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Bi₂S₃/Ti₃C₂-TPP nano-heterostructures induced by near-infrared for photodynamic therapy combined with photothermal therapy on hypoxic tumors

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Abstract

Background Photodynamic therapy (PDT) efficacy of bismuth sulfide (Bi₂S₃) semiconductor has been severely restricted by its electron–hole pairs (e[−]–h⁺) separation inefficiency and oxygen (O₂) deficiency in tumors, which greatly hinders reactive oxygen species (ROS) generation and further clinical application of Bi₂S₃ nanoparticles (NPs) in biomedicine.

Results Herein, novel Bi₂S₃/titanium carbide (Ti₃C₂) two-dimensional nano-heterostructures (NHs) are designed to realize multimode PDT of synchronous O₂ self-supply and ROS generation combined with highly efficient photothermal tumor elimination for hypoxic tumor therapy. Bi₂S₃/Ti₃C₂ NHs were synthesized via the in situ synthesis method starting from Ti₃C₂ nanosheets (NSs), a classical type of MXene nanostructure. Compared to simple Bi₂S₃ NPs, Bi₂S₃/Ti₃C₂ NHs significantly extend the absorption to the near-infrared (NIR) region and enhance the photocatalytic activity owing to the improved photogenerated carrier separation, where the hole on the valence band (VB) of Bi₂S₃ can react with water to supply O₂ for the electron on the Ti₃C₂ NSs to generate ·O₂[−] and ·OH through electron transfer. Furthermore, they also achieve ¹O₂ generation through energy transfer due to O₂ self-supply. After the modification of triphenylphosphium bromide (TPP) on Bi₂S₃/Ti₃C₂ NHs, systematic in vitro and in vivo evaluations were conducted, revealing that the synergistic-therapeutic outcome of this nanoplatform enables complete eradication of the U251 tumors without recurrence by NIR laser irradiation, and it can be used for computed tomography (CT) imaging because of the strong X-ray attenuation ability.

Conclusion This work expands the phototherapeutic effect of Bi₂S₃-based nanoplatforms, providing a new strategy for hypoxic tumor theranostics.

Keywords Nano-heterostructure, Photothermal therapy, Photodynamic therapy, Hypoxic tumor, Tumor theranostics

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Introduction

In recent years, nanomaterial (NM)-mediated phototherapy, including photodynamic therapy (PDT) and photothermal therapy (PTT), has shown great potential as an emerging method for oncotherapy, owing to its advantages of minimal invasiveness, regional selectivity, negligible toxic and side effects, short treatment time, and repeatable treatments [1]. At present, numerous NMs have been developed for PTT [2], but the progress of PDT is restricted due to the excitation of most photosensitizers (PSs) only by ultraviolet and visible light, the low electron–hole pair (e^-h^+) separation efficiency, and the reduced therapeutic effect caused by the massive consumption of tissue oxygen (O_2) during the treatment process [3]. Besides, tumor tissue inherently presents a hypoxic tumor microenvironment (TME) because of the distortional blood vessels, the rapid proliferation, and the exuberant metabolism of tumor cells [4]. In order to enhance the poor PDT effect attributed to hypoxia, various techniques of O_2 carrying and catalytic production of O_2 were investigated and applied [5], but the asynchronism between O_2 supply and reactive oxygen species (ROS) formation made the therapeutic effect on tumors not ideal. Although PTT is a non- O_2 -dependent therapeutic method [6], if photothermal agents (PTAs) are not uniformly dispersed in the distorted tumor vessels during treatment, the temperature rise would be uneven, which may also weaken the oncotherapy efficacy [7]. As a result, developing a simple nanoplatform that combines both PDT and PTT functions under near-infrared (NIR) irradiation at the same wavelength to achieve a synergistic therapeutic effect with the generation of ROS and heat while producing O_2 has greater clinical potential. Such a nanoplatform would have stronger ROS generation capacity and better therapeutic efficacy than nanomaterials known to have both PTT and PDT functions (e.g., indocyanine green), owing to its excellent oxygen supply capacity [8].

The ability of PSs to generate both ROS and O_2 simultaneously depends on their unique internal electronic structure. Photosensitive semiconductor NMs with relatively narrow bandgaps can be excited by NIR at 800–1300 nm to make the photoinduced electrons in the valence band (VB) migrate to the conduction band (CB), leaving positive holes [9]. These free-moving photoinduced electrons and holes with redox abilities can, respectively, react with the surrounding O_2 and water molecules under certain potential conditions to produce the killing superoxide anion ($\cdot O_2^-$) radicals and O_2 [10]. Moreover, as a primary ROS, $\cdot O_2^-$ can be further transformed into a series of ROS with stronger tumor-killing property, such as hydrogen peroxide (H_2O_2) and hydroxyl ($\cdot OH$) radicals [11]. $\cdot O_2^-$ radicals are produced through the type I PDT mechanism by the electron transfer

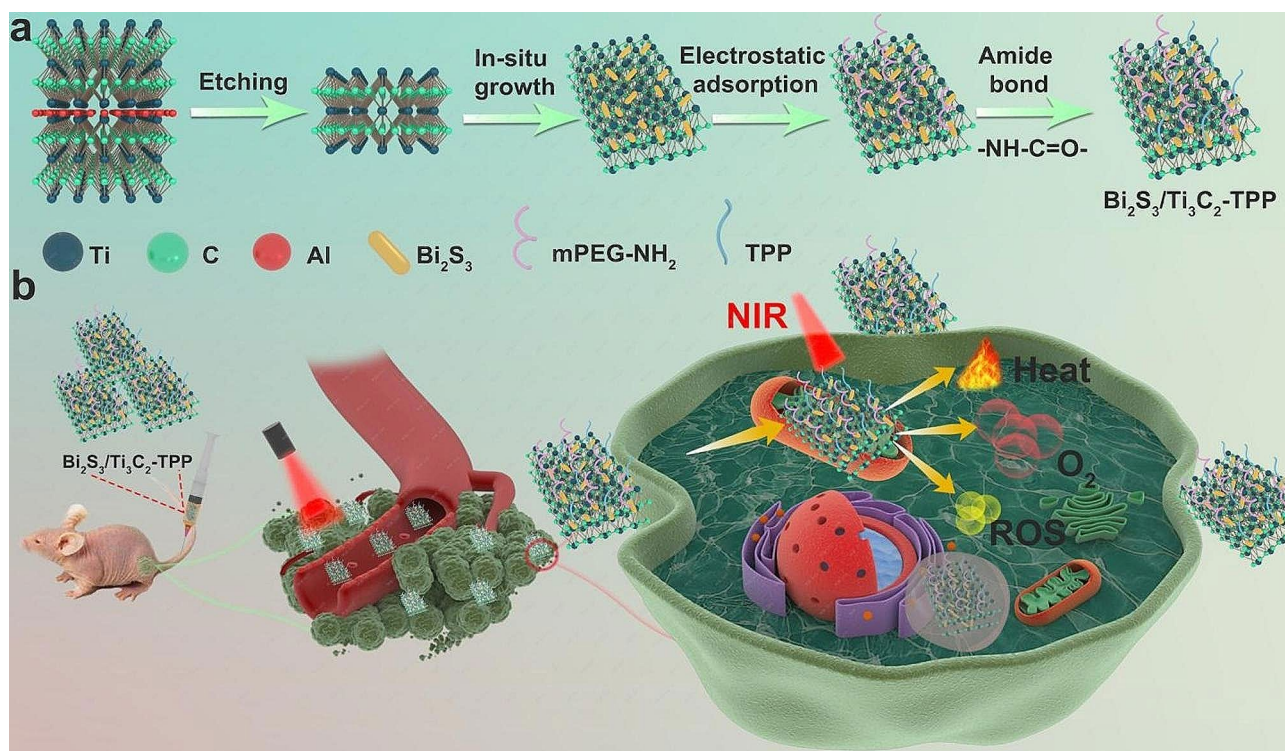
pathway, whereas most PSs are based on the type II PDT mechanism, in which energy transfers to O_2 in adjacent tissues to generate toxic singlet oxygen (1O_2) [12]. Bismuth sulfide (Bi_2S_3) nanoparticles (NPs) have excellent optical catalytic activity and a high NIR absorption coefficient, and have been widely utilized as PTAs [13] and computed tomography (CT) contrast agents [14] in recent decades to realize the integration of multi-modal diagnosis and treatment of tumors [15, 16], but their direct application in PDT is limited due to the narrow bandgap. Nevertheless, Bi_2S_3 -based nano-heterostructures (NHs) with appropriate wavelength absorption and boosted photogenerated carrier separation efficiency are promising for establishing PSs with novel properties. Regrettably, there has been little research in this field.

Herein, we designed a new nanoplatform, Bi_2S_3/Ti_3C_2 -TPP (TPP, triphenylphosphium bromide), that was synthesized via the in situ synthesis method starting from titanium carbide (Ti_3C_2), two-dimensional ultra-thin ceramic nanosheets (NSs) with satisfactory photothermal conversion efficiency (PTCE, η), excellent electrical features, and a large specific surface area [17], then further attached with TPP on the surface (Scheme 1). Compared to simple Bi_2S_3 NPs, Bi_2S_3/Ti_3C_2 -TPP significantly extends absorption to the NIR region, enhances photocatalytic activity due to higher photogenerated carrier separation and electron transfer efficiency, and can also effectively aggregate in tumor cells by targeting the mitochondria showing extraordinary CT imaging capability. Under 808 nm laser irradiation, Bi_2S_3 NPs generate free-moving holes in the VB and electrons in their CB that migrate to Ti_3C_2 NSs connected closely with Bi_2S_3 [18], during which $\cdot O_2^-$ and $\cdot OH$ were produced via the type I PDT mechanism by electron transfer and 1O_2 was generated via the type II PDT mechanism by energy transfer. Therefore, Bi_2S_3/Ti_3C_2 -TPP demonstrated powerful tumor diagnosis and treatment functions in both normoxic and hypoxic TME.

Materials and methods

Materials

HCl (36–38%), LiF, Ti_3AlC_2 (325-meshes), and TPAOH were supplied by Beijing Chemical Reagents Company (Beijing, China), Aladdin Bio-Chem Technology Corporation (Shanghai, China), Forsman Scientific Company (Beijing, China), and J&K Scientific Corporation (Beijing, China), respectively. Oleic acid (OA), oleylamine, thioacetamide, bismuth neodecanoate, polysorbate 20 (Tween 20), cyclohexane, mPEG-NH₂ ($M_w \approx 2K$), N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDC), N-hydroxysuccinimide (NHS), TPP, dimethyl sulfoxide (DMSO), 5,5-dimethyl-1-pyrroline N-oxide (DMPO), 2,2,6,6-tetramethylpiperidine (TEMP), CPZ, and colchicine were purchased from Sigma-Aldrich



Scheme 1 Schematic Illustration of the Fabrication Process and Therapeutic Mechanism of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$. **a** $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ was prepared by *in-situ* growth of Bi_2S_3 NPs in Ti_3C_2 NSs and modification with TPP. **b** $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ exerted a synergistic therapeutic effect of mitochondria-targeted multi-mode PDT combined with PTT on hypoxic tumors under NIR irradiation

(MO, USA). MTT, DCFH-DA, mitochondrial membrane potential assay kit (JC-1), cell lysis buffer (RIPA), Annexin V-FITC/propidium iodide (PI) apoptosis and necrosis detection kit, H&E staining kit, and TUNEL apoptosis assay kit were obtained from Beyotime Institute of Biotechnology (Shanghai, China). Calcein Acetoxymethyl ester (Calcein-AM)/PI double stain kit was brought from Yeasen Biotechnology Corporation (Shanghai, China). PBS (pH=7.4, 10 mM), fetal bovine serum (FBS), trypsin, penicillin-streptomycin (PS), reduced serum medium (Opti-MEM), DMEM, mitochondrion-selective probe (Mito-Tracker Red), and red hypoxia probe (Image-iT™) were provided by Gibco Life Technologies (CA, USA). All the polyclonal antibodies were obtained from Proteintech Group Incorporation (IL, USA). All chemicals were used as received without further purification.

Synthesis of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$

Ti_3C_2 NSs were prepared in a typical way of etching the Al atomic layer from Ti_3AlC_2 with HCl/LiF. Briefly, LiF (5 g) was dissolved in a HCl aqueous solution (50 mL, 9 mmol) contained in a polytetrafluoroethylene (PTFE) beaker and magnetically stirred for 30 min at room temperature to acquire a hydrogen fluoride (HF) aqueous solution. HF and HCl are both highly corrosive liquids, and the experimenters must strictly follow the safety

protocol during the experiment. Then, Ti_3AlC_2 powders (5 g) were slowly added in the HF aqueous solution and stirred at 38°C for 3 days. After being washed and centrifuged repeatedly with deionized water (DI) or ethanol (11,000 rpm \times 10 min) to make the supernatant pH close to 7, the collected precipitates were dispersed in 25 mL TPAOH with stirring at room temperature for 3 days. The resulting ultra-thin Ti_3C_2 NSs were collected by freeze-drying after washing and centrifugation with DI or ethanol several times (11,000 rpm \times 10 min). Particularly, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs were fabricated by a solvothermal synthetic method *in situ*. Initially, Ti_3C_2 NSs (0.25 g) were evenly dispersed in ethanol (10 mL) with an OA (20 mL) addition, and bismuth neodecanoate (1.45 g) was added to the mixed solution under continuous agitation, which made the bismuth ion (Bi^{3+}) attached to the Ti_3C_2 NSs by electrostatic absorption. Subsequently, an oleylamine solution (4 mL) with thioacetamide (0.15 g) dissolved in it was rapidly added to the above solution with violent stirring at room temperature for 1 h. Eventually, the prepared mixture was sealed in a Teflon-lined autoclave (50 mL) and heated at 150°C for 10 h, which was required to cool to room temperature naturally at the end of the reaction. After being washed and centrifuged with ethanol (10,000 rpm \times 5 min) three times, the oil-soluble OA-coated $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs were obtained by

freeze-drying. Similarly, Bi₂S₃ NPs were fabricated without the addition of Ti₃C₂ NSs.

Fabrication of Bi₂S₃/Ti₃C₂-TPP

In order to connect TPP to Bi₂S₃/Ti₃C₂ NHs, oil-soluble OA-coated Bi₂S₃/Ti₃C₂ NHs must be converted to water-soluble NHs and encapsulated with mPEG-NH₂. Firstly, OA-coated Bi₂S₃/Ti₃C₂ NHs (100 mg) and Tween 20 (150 μL) were uniformly dispersed in cyclohexane (20 mL) with stirring at room temperature for 1.5 h. After that, the mixture was added dropwise into DI (30 mL) in a 70 °C water bath and continuously stirred for at least 3 h to evaporate the cyclohexane. Later, the water-soluble Tween 20-functionalized Bi₂S₃/Ti₃C₂ NHs were collected by freeze-drying after washing and centrifugation with ethanol (10,000 rpm × 5 min) three times. Then, mPEG-NH₂ (20 mg) and Tween 20-functionalized Bi₂S₃/Ti₃C₂ NHs (20 mg) were sufficiently dispersed in DI (10 mL) and stirred overnight at room temperature, while EDC (50 mg) and NHS (25 mg) were added into an aqueous solution (10 mL) containing TPP (50 mg) and stirred in the dark at room temperature for 6 h to activate TPP. Finally, the two prepared solutions were mixed with stirring overnight in the dark, and Bi₂S₃/Ti₃C₂-TPP was obtained by freeze-drying after washing and centrifugation with DI (10,000 rpm × 5 min) three times.

Characterization

SEM (1000000X, TESCAN, CZ), TEM (JEM-2100, JEOL, JPN), and AFM (Dimension Fastscan, BRUKER, GER) were used to describe the morphology and structure of the samples. XRD (X'PERT, Panalytical, NL), XPS (ESCALAB 250Xi, ThermoFisher, USA), and FT-IR (Nicolet is50, ThermoFisher, USA) were conducted to determine the composition of the samples. ICP-MS, PL, UV-vis, and ESR spectroscopy were collected on an iCAP7400 spectrometer (ThermoFisher, USA), a FLS1000 spectrophotometer (Edinburgh, UK), a LAMBDA950 spectrophotometer (PE, USA), and an A200S-95/12 spectrometer (BRUKER, GER), respectively. The average size and zeta potential of the samples were measured by Nano-MS2000/ZS90 (Malvern, UK). The NIR source was purchased from LeiShi (808 nm, China). Oxygen concentration and temperature were severally monitored by a dissolved oxygen meter (JPBJ-607 A, INESA, China) and a portable thermal imager (E5-XT, FLIR, USA). All fluorescent imaging was visualized by an inverted fluorescence microscope (AF6000, Leica, GER). Cell viability was evaluated by a microplate reader (ELX800, Bio Tek, USA), while a flow cytometer (NovoCyte, Agilent, USA) was used to conduct quantitative flow cytometry. Western blot exposure was performed on a chemiluminescence imager (ChemiDoc Touch, Bio-Rad, USA).

Photodynamic performance and O₂ evolution evaluation

ROS assay kits were used to detect the ROS generation capacity of Bi₂S₃/Ti₃C₂, Ti₃C₂, and Bi₂S₃ under 808 nm laser irradiation. Initially, the DCFH-DA was diluted with ethanol to acquire a solution with a molar concentration of 1 mM, and the mixture (0.5 mL) was added to the NaOH solution (2 mL, 10 mM) with stirring at room temperature for 30 min in the dark. After that, the obtained mixed solution was uniformly dispersed in PBS (10 mL, pH=7.4, 10 mM) and stored at 4 °C in a refrigerator. Then, the prepared DCFH-DA solution (100 μL) was mixed with the sample solution (100 μL, 200 μg mL⁻¹) and irradiated with an 808 nm laser (1 W cm⁻², 10 min). Meanwhile, a PL spectrophotometer was used to measure the fluorescence intensity of DCF per minute in order to determine how ROS generation changed over time.

ESR measurements were performed to identify the ROS species generated by Bi₂S₃/Ti₃C₂ under NIR irradiation. Briefly, TEMP as a spin-trapping agent was added to the methanol solution of Bi₂S₃/Ti₃C₂ (500 μL, 100 μg mL⁻¹) to reach a molar concentration of 200 μmol L⁻¹, and the mixture was irradiated with an 808 nm laser (1 W cm⁻², 10 min). Whereafter, an ESR measurement was immediately conducted to determine whether ¹O₂ was formed. The formation of ·OH and ·O₂⁻ was tested by the same experimental procedure in DI and methanol, respectively, and DMPO was used as the spin-trapping agent.

Nitrogen (N₂) was blown into the Bi₂S₃/Ti₃C₂ solution (10 mL, 100 μg mL⁻¹) for 15 min to remove the inherent O₂ in it, and the solution was sealed with liquid paraffin. After that, the amount of O₂ produced by the samples under 808 nm laser irradiation (1 W cm⁻²) was measured per minute within 20 min, utilizing a dissolved oxygen meter.

Photothermal performance evaluation

To evaluate the photothermal effect of the samples, Bi₂S₃/Ti₃C₂, Ti₃C₂, or Bi₂S₃ aqueous solutions (4.5 mL, 100 μg mL⁻¹) were added to a tailor-made quartz cuvette and irradiated with an 808 nm laser (1 W cm⁻², 10 min). A thermal imager was used to record temperature changes every 30 s, until it reached room temperature approximately. Simultaneously, the above experimental procedure was continuously repeated five times to further investigate the photothermal stability of Bi₂S₃/Ti₃C₂. Subsequently, the temperature changes of Bi₂S₃/Ti₃C₂ aqueous solutions with different concentrations (0, 25, 50, 100, and 200 μg mL⁻¹) were recorded within 10 min under the same conditions, aiming to determine the influence of sample concentration on the photothermal performance.

Enrichment analysis of hypoxic-related genes in glioma

Firstly, gene expression and clinical data from glioma samples were gathered through the TCGA and CGGA databases. Then, the hypoxic-related gene set was chosen from The Molecular Signatures Database (MSigDB, <http://www.gsea-msigdb.org/gsea/msigdb/human/search.jsp>). The samples were classified into low-grade glioma and glioblastoma based on tumor grade, and the enrichment degree of hypoxic-related genes in the two samples was determined by GSEA. Whereafter, the R package of gene set variation analysis (GSVA) and ssGSEA were used to acquire hypoxic-related functional scores for each glioma patient sample in the TCGA and CGGA databases. Finally, the scores were analyzed in conjunction with tumor grade and survival time to explore their relationship.

Cellular experiments

U251 human glioma cells were selected for the following cellular investigations and cultured according to the standard protocol. For cell culture in a hypoxic microenvironment, the conditions were set as a humidified atmosphere containing 1% O₂ and 5% CO₂ at 37 °C, which was formed by flowing N₂.

For cellular uptake, 8 × 10⁴ U251 cells in 1 mL of DMEM were seeded in each well of a 12-well plate for 24 h of incubation. Then, cells were treated with Opti-MEM containing Bi₂S₃/Ti₃C₂-TPP or Bi₂S₃/Ti₃C₂ (12.5, 25, 50, 100, or 200 μg mL⁻¹) for 4 h. After washing with PBS three times, cells were digested with trypsin and collected by centrifugation (1000 rpm × 5 min, 4 °C). Finally, cells were counted, and the Bi or Ti content was measured by ICP-MS tests.

For the hemolysis assay, 0.5 mL of blood was obtained from mice, and erythrocytes were extracted by centrifuging and washing with PBS five times (3000 rpm × 5 min, 4 °C) and diluted at 10 times their initial volume in PBS. Then, diluted suspensions (200 μL) were mixed with 800 μL of PBS, Triton-100 (0.025% in PBS), and Bi₂S₃/Ti₃C₂-TPP (25, 50, 100, 200, and 300 μg mL⁻¹ in PBS) and incubated with cells at 37 °C for 2 h. All samples were centrifuged (3000 rpm × 5 min, 4 °C), and the supernatant absorbance at 541 nm was detected using a microplate reader. The hemolysis rate was calculated by dividing the supernatant absorbance value (I) of the samples by the value (I₀) of the positive control group. The calculation formula is as follows:

$$\text{Hemolysis rate (\%)} = (I/I_0) \times 100\%$$

For mitochondrial targeting performance, 1.6 × 10⁵ U251 cells in 2 mL DMEM were seeded in dishes (d=35 mm) for 24 h of incubation. After treatment with Opti-MEM containing Bi₂S₃/Ti₃C₂-TPP (100 μg mL⁻¹) for 4 h, Mito-Tracker Red (1 μL, 400 nmol L⁻¹) was added to the medium, and cells were incubated for

another 15 min. Finally, cells were washed with PBS three times and imaged with an inverted fluorescence microscope.

For mitochondrial membrane potential assessment, 8 × 10⁴ U251 cells in 1 mL of DMEM were seeded in each well of a 12-well plate for 24 h of incubation. After treatment with Opti-MEM containing different NMs (100 μg mL⁻¹) for 4 h, cells were irradiated with or without an 808 nm laser (1 W cm⁻², 10 min). After another 2 h of incubation, cells were stained with JC-1 (5 μmol L⁻¹, 20 min) in Opti-MEM for mitochondrial membrane potential. Finally, cells were washed with PBS three times and directly observed by an inverted fluorescence microscope or quantitatively analyzed by flow cytometry.

For intracellular O₂ detection, 1 × 10⁴ U251 cells in 100 μL of DMEM were seeded in each well of a 96-well plate for 24 h of incubation. After treatment with Opti-MEM containing Bi₂S₃/Ti₃C₂-TPP (100 μg mL⁻¹) for 4 h, Image-iT™ was added to the medium to a concentration of 10 μmol L⁻¹, and cells were incubated for another 30 min. Then, the medium was renewed, and cells were incubated in a hypoxic culture environment for 4 h. After the incubation, cells were irradiated with or without an 808 nm laser (1 W cm⁻², 10 min) and imaged with an inverted fluorescence microscope.

For intracellular ROS detection, 8 × 10⁴ U251 cells in 1 mL of DMEM were seeded in each well of a 12-well plate for 24 h of incubation. After treatment with Opti-MEM containing Bi₂S₃/Ti₃C₂-TPP (100 μg mL⁻¹) for 4 h, cells were stained with DCFH-DA (10 μmol L⁻¹) in Opti-MEM for 15 min and irradiated with or without an 808 nm laser (1 W cm⁻², 10 min). Finally, the cellular DCF fluorescence was directly observed by an inverted fluorescence microscope or quantitatively analyzed by flow cytometry after washing with PBS three times.

For in vitro phototherapy, 1 × 10⁴ U251 cells in 100 μL DMEM were seeded in each well of a 96-well plate for 24 h of incubation. After treatment with Opti-MEM containing different NMs (100 μg mL⁻¹) for 4 h, cells were irradiated with or without an 808 nm laser (1 W cm⁻², 10 min) and incubated for another 20 h. Then, cell viability was evaluated by MTT and live/dead staining assays. For PDT fractions, the temperature during NIR irradiation should be 4 °C to avoid the influence of PTT.

For the cell clone formation assay, 1 × 10³ U251 cells in 2 mL DMEM were seeded in each well of a 6-well plate for 2 d of incubation. After treatment with Opti-MEM containing different NMs (100 μg mL⁻¹) for 4 h, cells were irradiated with or without an 808 nm laser (1 W cm⁻², 10 min) and incubated for another 10 d with medium changed every other day. When significant clone clusters were visible under the microscope, the incubation was terminated. Then, cells were washed with PBS and fixed with formaldehyde for 15 min. Finally,

0.1% crystal violet was added to the samples to stain for 10 min, which was photographed for recording.

For the cell apoptosis assay, 1.6×10^5 U251 cells in 2 mL DMEM were seeded in each well of a 6-well plate for 24 h of incubation. After treatment with Opti-MEM containing different NMs ($100 \mu\text{g mL}^{-1}$) for 4 h, cells were irradiated with or without an 808 nm laser (1 W cm^{-2} , 10 min) and incubated for another 20 h. Then, cells were washed with cold PBS, collected, and stained with the Annexin V-FITC/PI assay kit. Finally, they were detected by flow cytometry.

For Nrf2, HO-1, and Hsp70 expressions, 1.6×10^5 U251 cells in 2 mL DMEM were seeded in each well of a 6-well plate for 24 h of incubation. After treatment with Opti-MEM containing different NMs ($100 \mu\text{g mL}^{-1}$) for 4 h, cells were irradiated with or without an 808 nm laser (1 W cm^{-2} , 10 min) and incubated for another 20 h. Then, cells were washed with cold PBS and collected. The protein content was determined by the Bradford method, and the expressions of Nrf2, HO-1, Hsp70, and β -actin were evaluated through western blotting analysis.

Animal experiments

Four-week-old female Balb/c nude mice were purchased from Liaoning Changsheng Biotechnology Co., Ltd., and all animal experiments were approved by the Animal Care and Ethical Committee of the First Affiliated Hospital of Harbin Medical University, which were conducted in accordance with the guidelines from the Ministry of Science and Technology of China.

For in vivo phototherapy, the U251 tumor model was established by subcutaneous injection of PBS ($100 \mu\text{L}$) containing 1×10^6 U251 cells into the right back of the hips of each mouse. When the tumor volume of mice reached approximately 80 mm^3 ($V = \text{width}^2 \times \text{length} \times 1/2$) under the same feeding conditions, total mice were randomly divided into 7 treatment groups ($n=5$) and injected through the caudal vein with (I) PBS (pH=7.4, 10 mM); (II) PBS (pH=7.4, 10 mM) with an 808 nm laser irradiation (1 W cm^{-2} , 10 min); (III) $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP; (IV) Bi_2S_3 with an 808 nm laser irradiation (1 W cm^{-2} , 10 min); (V) $\text{Bi}_2\text{S}_3 + \text{Ti}_3\text{C}_2$ with an 808 nm laser irradiation (1 W cm^{-2} , 10 min); (VI) $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ with an 808 nm laser irradiation (1 W cm^{-2} , 10 min); (VII) $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP with an 808 nm laser irradiation (1 W cm^{-2} , 10 min), respectively. The injection concentration of the samples contained in PBS ($200 \mu\text{L}$) was 20 mg kg^{-1} , with a Bi_2S_3 to Ti_3C_2 ratio of 3:2 in Group V. And the samples were irradiated at 24 h post-injection. Under NIR irradiation, the temperature of tumor sites in mice was recorded by a portable thermal imager. After treatments, the tumor volumes and body weights of mice were monitored every other day during 2 weeks. Furthermore, the mice were euthanized after 2-week tumor treatment, and the tumor

tissues and major organs (heart, liver, spleen, lung, and kidney) of mice in each group were harvested, sliced, and stained for H&E and TUNEL staining to perform the histological analysis.

For in vivo biodistribution and optical imaging assessment, U251 tumor-bearing mice were intravenously injected with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP (20 mg kg^{-1}). At indicated time points (1, 2, 4, 8, 12, and 24 h post-injection), $50 \mu\text{L}$ blood was extracted from the caudal vein of each mouse and weighed every time ($n=5$ for each time point). The mice were sacrificed after the last blood collection, and tumors and major organs were harvested. Then, the content of Bi or Ti in the blood and tissue samples was measured by ICP-MS tests. In vivo optical imaging was presented by CT imaging. The mouse was anesthetized at 0 and 8 h post-injection, and imaged by a CT scanner. Then, the CT value of the region of interest (ROI) at two time points was evaluated and compared with the corresponding analysis software.

Statistical analysis

SPSS 23.0 statistical software was conducted for the statistical analysis of the experimental data. A completely random design was performed to obtain measurement data. The normal distribution was described by mean \pm standard deviation, and the non-normal distribution was described by quartile and median. The analysis of variance (ANOVA) was used for comparison between multiple groups, and the t test was used for comparison between two groups. $P < 0.05$ was considered statistically significant.

Results and discussion

Fabrication and characterization of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP

Two-dimensional Ti_3C_2 NSs were synthesized by the typical method of liquid-phase chemical etching and exfoliation described in the previous literature [19, 20]. Firstly, titanium aluminum carbide (Ti_3AlC_2) powder, the MAX phase precursor, was etched with a relatively mild mixed aqueous solution of lithium fluoride (LiF) and hydrogen chloride (HCl) to remove the Al layers, yielding well-stacked and uniform-thickness multilayer Ti_3C_2 sheets. Then, in order to acquire ultrathin Ti_3C_2 NSs, the multilayer Ti_3C_2 sheets were further delaminated by tetrapropylammonium hydroxide (TPAOH) intercalation to increase layer spacing and adequately dispersed by ultrasound to reduce their planar sizes. The microstructures of multilayer massive Ti_3C_2 formed after the etching of Ti_3AlC_2 powder and ultrathin Ti_3C_2 NSs obtained after exfoliation and ultrasonic dispersion were clearly visible in scanning electron microscopy (SEM), transmission electron microscopy (TEM), and atomic force microscopy (AFM) images. According to the SEM image, Ti_3C_2 sheets had multilayer stacked structures with

uniform thickness (Fig. 1a). The TEM image showed that the average planar size of virtually transparent ultrathin Ti_3C_2 NSs was about 100 nm (Fig. 1b), while the AFM images indicated that the mean thickness was about 1 nm (Fig. 1c, d). Bi_2S_3 NPs were synthesized in situ on the surface of ultrathin Ti_3C_2 NSs by the solvothermal synthesis method to obtain oil-soluble $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs [21, 22]. The TEM image revealed that Bi_2S_3 NPs with approximately 65 nm in length and 15 nm in diameter were randomly loaded on the exterior of Ti_3C_2 NSs (Fig. 1e), and two fringes with interplanar distances of 0.310 nm and 0.231 nm were matched with the (2 1 1) plane of Bi_2S_3 and (1 0 4) plane of Ti_3C_2 , respectively, according to the high-resolution TEM (HRTEM) image (Fig. 1f). The distribution of elements in $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs was presented by the element mapping of energy dispersive X-ray spectroscopy (EDS) (Fig. 1g). All of these morphologically confirmed the synthesis of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs. The X-ray

diffraction (XRD) patterns demonstrated that the typical diffraction peaks of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs could be indexed to the orthorhombic Bi_2S_3 phase (JCPDS No. 75-1306) and the hexagonal Ti_3C_2 phase (JCPDS No. 52-0875) (Fig. 1h and Additional file 1: Fig. S1), and the characteristic peaks of Bi_2S_3 and Ti_3C_2 were visible in the X-ray photoelectron spectroscopy (XPS) spectra of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs (Fig. 1i, j and Additional file 1: Fig. S2). These two findings further verified the successful synthesis of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs. The ultraviolet-visible (UV-vis) spectrum indicated that the absorption intensity of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs in the NIR region was significantly higher than that of Bi_2S_3 NPs (Fig. 1k), which would be conducive to the improvement of PDT efficiency. The bandgap of Bi_2S_3 was calculated to be about 1.43 eV consistent with the previous report by the formula: $(\alpha h\nu)^n = m(h\nu - E_g)$ (Additional file 1: Fig. S3) [23], where α , h , ν and E_g are absorption coefficient, Planck constant, photon frequency and bandgap energy,

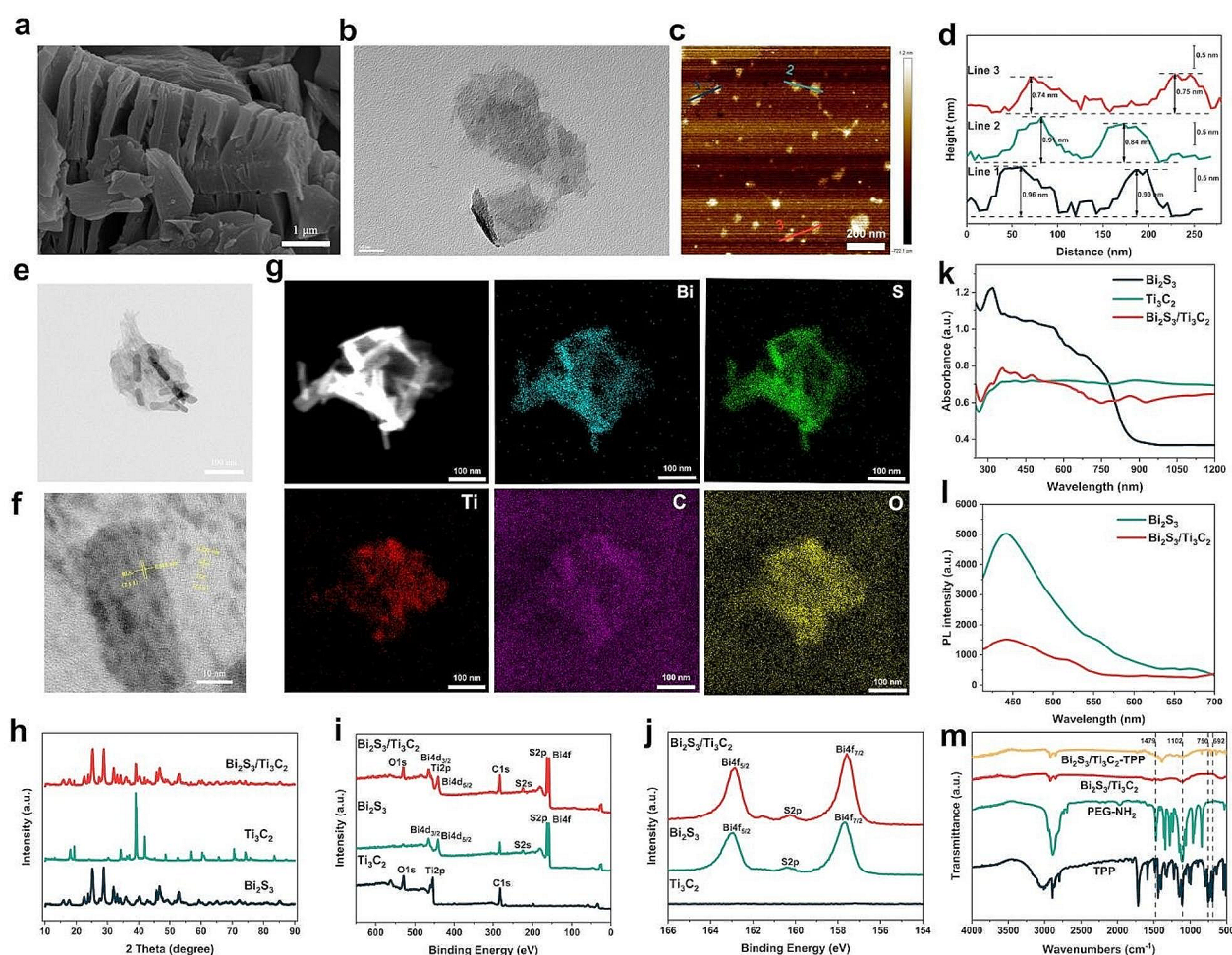


Fig. 1 Fabrication and characterization of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP. **a** SEM image of Ti_3C_2 powders. **b** TEM image of Ti_3C_2 NSs. **c-d** AFM images of Ti_3C_2 NSs. **e** TEM image, **f** HRTEM image, and **g** EDS mapping images (Bi-blue, S-green, Ti-red, C-purple, and O-yellow) of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs. **h** XRD patterns, **i-j** XPS spectra, and **k** UV-vis absorption spectra of Bi_2S_3 NPs, Ti_3C_2 NSs, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs. **l** PL spectra of Bi_2S_3 NPs and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs. **m** FT-IR patterns of TPP, PEG-NH₂, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP

respectively, and n , m are constants. It satisfied the energy requirement for an 808 nm laser to excite NMs with a bandgap < 1.53 eV [24]. Photoluminescence (PL) effect resulting from the combination of electrons and holes generated by photoexcitation could be used to evaluate the e^-h^+ separation ability of PSs. The PL spectrum presented that the fluorescence intensity of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs was substantially reduced compared to Bi_2S_3 NPs (Fig. 11), with the greatest attenuation in the preparation ratio of 3:2 (Additional file 1: Fig. S4), suggesting that the e^-h^+ separation efficiency of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs was notably enhanced and useful for the further promotion of PDT efficiency.

To improve the biocompatibility and achieve the tumor-targeting capability of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs, oil-soluble $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs were converted to water-soluble NHs, which were then encapsulated with methoxypropylene glycol amine (mPEG-NH₂) by electrostatic adsorption and attached with TPP on the surface via an amide bond (-NH-C=O-). Compared with the Fourier transform infrared (FT-IR) spectrum of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$, the vibration of C-O-C at 1102 cm⁻² from mPEG-NH₂ and the characteristic absorption peaks consistent with the molecular structure of TPP, including benzene skeleton vibration at 1479 cm⁻² and benzene ring monosubstituted peak vibration that appeared simultaneously at 750 cm⁻² and 692 cm⁻², were visible in the FT-IR spectrum of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP (Fig. 1m). This analysis proved that mPEG-NH₂ and TPP had been successfully conjugated on $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs. Dynamic light scattering (DLS) analysis presented the change in hydration particle size during each step of the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP synthesis process (Additional file 1: Fig. S5a). The zeta potential measurements could also be utilized to represent the functionalized process of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs (Additional file 1: Fig. S5b). The zeta potential of Ti_3C_2 NSs changed from -27.2 mV \pm 0.6 mV to 3.8 mV \pm 0.8 mV after loading Bi_2S_3 NPs. The next encapsulation with mPEG-NH₂ and attachment with TPP made the zeta potential change to -11.7 mV \pm 1.0 mV and -8.0 mV \pm 0.2 mV, respectively. These results confirmed the completion of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP fabrication. Furthermore, DLS analysis showed that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP had good stability in pure water or Dulbecco's modified eagle medium (DMEM), which would be beneficial for its application in biomedical research (Additional file 1: Fig. S5c and Fig. S6). Finally, the UV-vis spectra further verified that TPP was properly bonded to $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs (Additional file 1: Fig. S7a), and the standard curve of TPP aqueous solution was obtained by measuring UV-vis absorption spectra at different concentrations (Additional file 1: Fig. S7b, c), which was performed to calculate that the connection rate of TPP on $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs was 42.35%.

Photodynamic and photothermal performance of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$

Under NIR irradiation, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs generated ROS through the synergistic function of type I and type II PDT mechanisms and produced O₂ simultaneously to enhance their therapeutic effect, as shown in Fig. 2a. Due to the fact that the Fermi energy level (E_F) of Ti_3C_2 was lower than that of closely connected Bi_2S_3 in the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs system [18, 25], a Schottky junction with a built-in electric field (E-field) could be formed at the contact interface of Ti_3C_2 and Bi_2S_3 . According to the Mott-Schottky plots, the CB potential of Bi_2S_3 was about -0.15 V vs. NHE (Additional file 1: Fig. S8), whereas the E_F of the O-terminated Ti_3C_2 was previously reported to be 0.71 eV vs. NHE [18, 26]. Thus, under 808 nm laser irradiation, the electrons of Bi_2S_3 were excited from VB to CB, resulting in activated electrons and holes, and the electrons were accelerated to transfer to Ti_3C_2 with an excellent conductivity property under the formation of an E-field that could effectively prevent their backflow to Bi_2S_3 [27]. This electron transfer process enhanced the separation efficiency of photoinduced electrons and holes in Bi_2S_3 , as well as endowed $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs with strong redox capacity. The activated electrons could react with O₂ to generate $\cdot\text{O}_2^-$ (Additional file S1: Eq. 1), while the activated holes could react with H₂O to produce O₂ (Additional file S1: Eq. 2). However, $\cdot\text{O}_2^-$, as a primary ROS, could be further transformed into $\cdot\text{OH}$ with a stronger killing effect (Additional file S1: Eqs. 3, 4). The above ROS were generated through the electron transfer pathway of the type I PDT mechanism, and the O₂ produced during the process could be converted into another cytotoxic ROS, $^1\text{O}_2$, through the energy transfer pathway of the type II PDT mechanism. Therefore, under 808 nm laser irradiation, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs could fulfill the synergistic effect of type I and type II PDT with or without abundant O₂ to enhance PDT efficacy.

To emphasize the superiority of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs, their abilities to generate O₂ and ROS under NIR irradiation were explored. It was shown that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs produced more O₂ than Bi_2S_3 NPs under 808 nm laser irradiation (1 W cm⁻²) for 20 min by measuring the O₂ content in sample solutions with a dissolved oxygen meter (Fig. 2b). The production of O₂ could not only facilitate type II PDT, but also alleviate TME hypoxia. Whereafter, using a ROS assay kit in which non-fluorescent 2, 7-dichloro-dihydrofluorescein diacetate (DCFH-DA) could be oxidized by ROS to form fluorescent 2, 7-dichlorofluorescein diacetate (DCF), we found that the DCF fluorescence intensity in $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ aqueous solution elevated progressively within 10 min under 808 nm laser irradiation (1 W cm⁻²), indicating the continuous generation of ROS (Fig. 2c). In addition, the DCF fluorescence intensity in $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ aqueous solution was

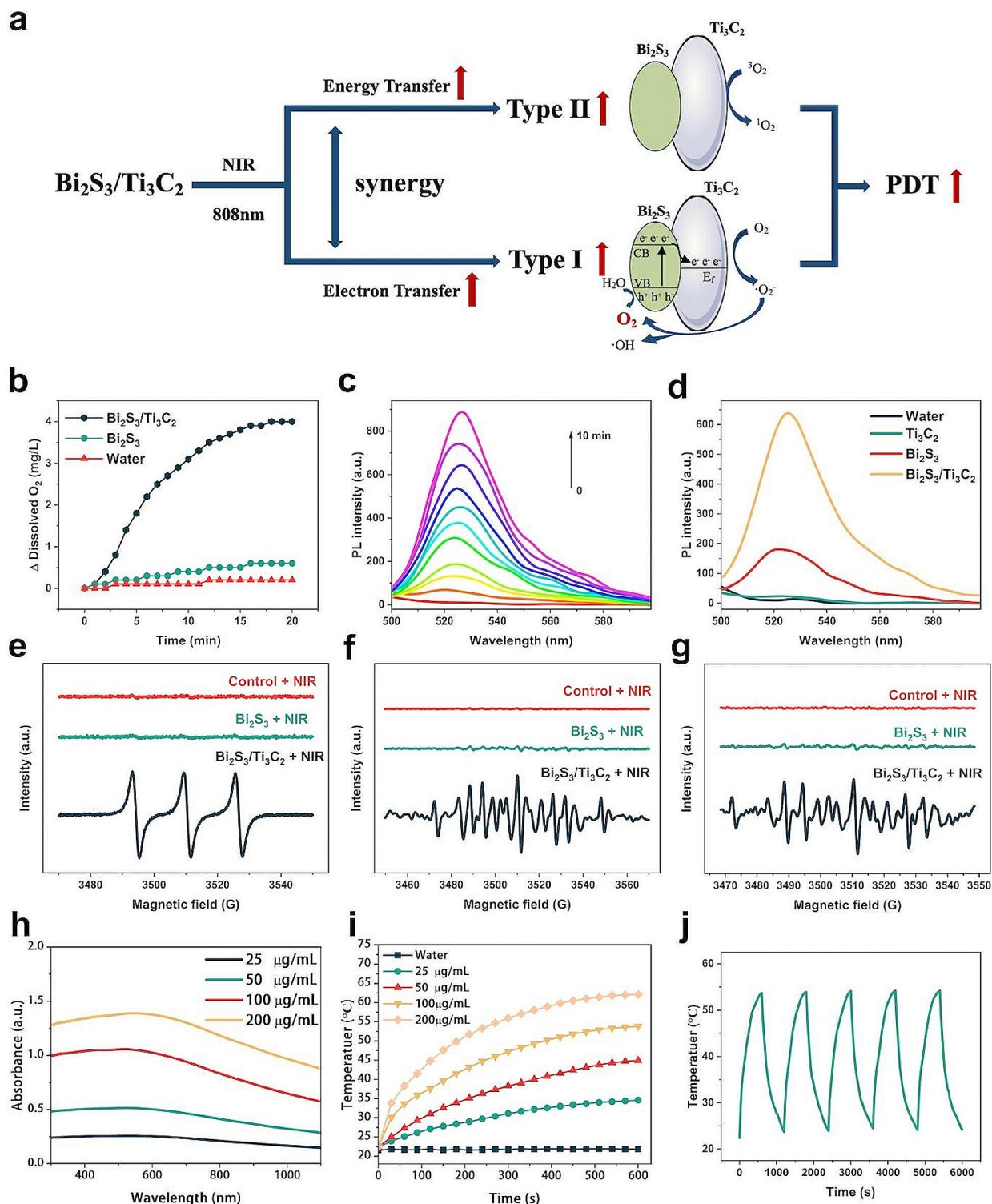


Fig. 2 PDT mechanism of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ and its photodynamic and photothermal performance. **a** Illustration of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ as PSs for enhanced PDT through the synergetic effect of type I and type II PDT under NIR irradiation. **b** O_2 production of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ and Bi_2S_3 aqueous solutions ($100 \mu\text{g mL}^{-1}$) under 808 nm laser irradiation (1 W cm^{-2}). **c** Time-dependent ROS generation by $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ ($100 \mu\text{g mL}^{-1}$) detected by DCFH-DA probe under 808 nm laser irradiation (1 W cm^{-2}). **d** ROS generation of Ti_3C_2 , Bi_2S_3 , and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ ($100 \mu\text{g mL}^{-1}$) under 808 nm laser irradiation (1 W cm^{-2}) detected by DCFH-DA probe. ESR spectra of Bi_2S_3 and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ under 808 nm laser irradiation (1 W cm^{-2}) for characterizing the generation of **e** $\cdot\text{O}_2^-$, **f** $\cdot\text{O}_2^-$, and **g** $\cdot\text{OH}$. **h** The UV-vis absorption spectra of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ with various concentrations. **i** Photothermal heating curves of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ with various concentrations under 808 nm laser irradiation (1 W cm^{-2}). **j** Photostability of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ ($100 \mu\text{g mL}^{-1}$) under 808 nm laser irradiation (1 W cm^{-2})

significantly stronger than that in both Bi_2S_3 NPs and Ti_3C_2 NSs aqueous solutions under 808 nm laser irradiation (1 W cm^{-2}) (Fig. 2d), which verified that the photocatalytic efficiency of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs was notably improved over the other two, since the formation of the Schottky junction with the existence of an E-field via the close integration of Bi_2S_3 and Ti_3C_2 could not only increase the rate of electron transfer but also prevent the combination of electrons and holes, thereby elevating the e^-h^+ separation greatly [28]. Subsequently, the species of generated ROS were identified by electron spin-resonance (ESR) spectroscopy. In accordance with the detection results, the signal intensity of $^1\text{O}_2$ in $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs aqueous solution was much higher than in Bi_2S_3 NPs aqueous solution under 808 nm laser irradiation (1 W cm^{-2} , 10 min) (Fig. 2e), because $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs had the ability to produce O_2 under NIR irradiation, which could increase the amount of $^1\text{O}_2$ generated by energy transfer. Similarly, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs aqueous solution exhibited strong $\cdot\text{O}_2^-$ (Fig. 2f) and $\cdot\text{OH}$ (Fig. 2g) signals under the same irradiation conditions, whereas almost no signals were observed in Bi_2S_3 NPs aqueous solution. These findings further confirmed that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs could raise the production of $^1\text{O}_2$ through the O_2 -dependent type II PDT mechanism and generate $\cdot\text{O}_2^-$ and $\cdot\text{OH}$ through the non- O_2 -dependent type I PDT mechanism.

$\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs had strong absorption in the NIR region, and their extinction coefficient was obviously higher than that of Bi_2S_3 NPs and Ti_3C_2 NSs (Fig. 2h and Additional file 1: Fig. S9), indicating that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs could exhibit more favorable photothermal performance, which was investigated under NIR irradiation. It was revealed that the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ aqueous solution manifested a concentration-dependent temperature rise curve by irradiating different concentrations of sample solutions with an 808 nm laser (1 W cm^{-2} , 10 min) (Fig. 2i). Among them, the temperature of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ aqueous solution with a concentration of $200 \mu\text{g mL}^{-1}$ increased to $62.1 \text{ }^\circ\text{C}$ after NIR irradiation, while the temperature of pure water was only $21.9 \text{ }^\circ\text{C}$ under the same irradiation condition, demonstrating that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ aqueous solution had a remarkable photothermal effect. Based on the maximum steady-state temperature and the time constant for heat transfer acquired from the cooling curve, the PTCE of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs was calculated to be 42.13% higher than that of Bi_2S_3 NPs (33.37%) and Ti_3C_2 NSs (29.89%) (Additional file 1: Fig. S10), which was also superior to those representative NMs with good photothermal effect in the NIR region, such as Au nanorods (21%) [29], MoS_2 NSs (24.37%) [30], Cu_9S_5 nanocrystals (25.7%) [31], and Au nanovesicles (37%) [32]. Moreover, the temperature change of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs aqueous solution was recorded under 808 nm laser irradiation (1 W cm^{-2}) for 10 min (laser on) followed by natural cooling to

room temperature (laser off), which was repeated for five cycles (Fig. 2j). The obtained profiles were generally consistent, suggesting that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs had satisfactory photothermal stability and the potential to be employed for durable photothermal treatment of tumors. All of the above results revealed that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs had excellent photothermal performance.

Selection of a hypoxic tumor model

It was demonstrated that hypoxic-related genes had considerable functional enrichment in glioblastoma by enrichment analysis of the hypoxic-related gene set, HALLMARK_HYPOXIA.v2023.1.Hs.gmt, in The Cancer Genome Atlas (TCGA, <https://portal.gdc.cancer.gov/>) and Chinese Glioma Genome Atlas (CGGA, <http://www.cgga.org.cn/>) databases (Fig. 3a, b). Then, in order to better understand the enrichment of the hypoxic-related gene set in each glioma patient sample, we performed hypoxic-related functional scores on the samples in the TCGA and CGGA by single-sample gene set enrichment analysis (ssGSEA), which was further analyzed in conjunction with the clinical data. The results (Fig. 3c, d) revealed that hypoxia enrichment scores on the samples increased with tumor grade progression in the two databases. Among them, glioblastoma, the most aggressive glioma, had the highest hypoxia enrichment score. Finally, the joint analysis of the relationship between hypoxia enrichment score and prognosis time of glioma patients found that an increase in hypoxia enrichment score was negatively correlated with patient survival time (Fig. 3e, f). These findings indicated that hypoxia had become an essential element in the malignant progression of glioma. Thus, changing the hypoxic status in the TME of glioma might be an important means of delaying tumor progression, which is why glioma was chosen for the following biological experiments.

Cytotoxicity and cellular uptake of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP

The cytotoxicity of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP was assessed by 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2 H-tetrazolium bromide (MTT) assays. The results showed that the cell viabilities of human glioma (U251) cells, human glioblastoma (A172) cells, and human umbilical vein endothelial (HUVEC) cells remained around 90%, even if the concentration of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP was as high as $300 \mu\text{g mL}^{-1}$ (Fig. 3g), indicating that the toxicity of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP to the above cells was negligible. In addition, for future biomedical research, the compatibility of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP on mouse red blood cells was evaluated by hemolysis assays, and it was found that when the concentration of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP was increased to $300 \mu\text{g mL}^{-1}$, the hemolysis rate was still less than 4% (Fig. 3h), implying that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP had credible hemocompatibility.

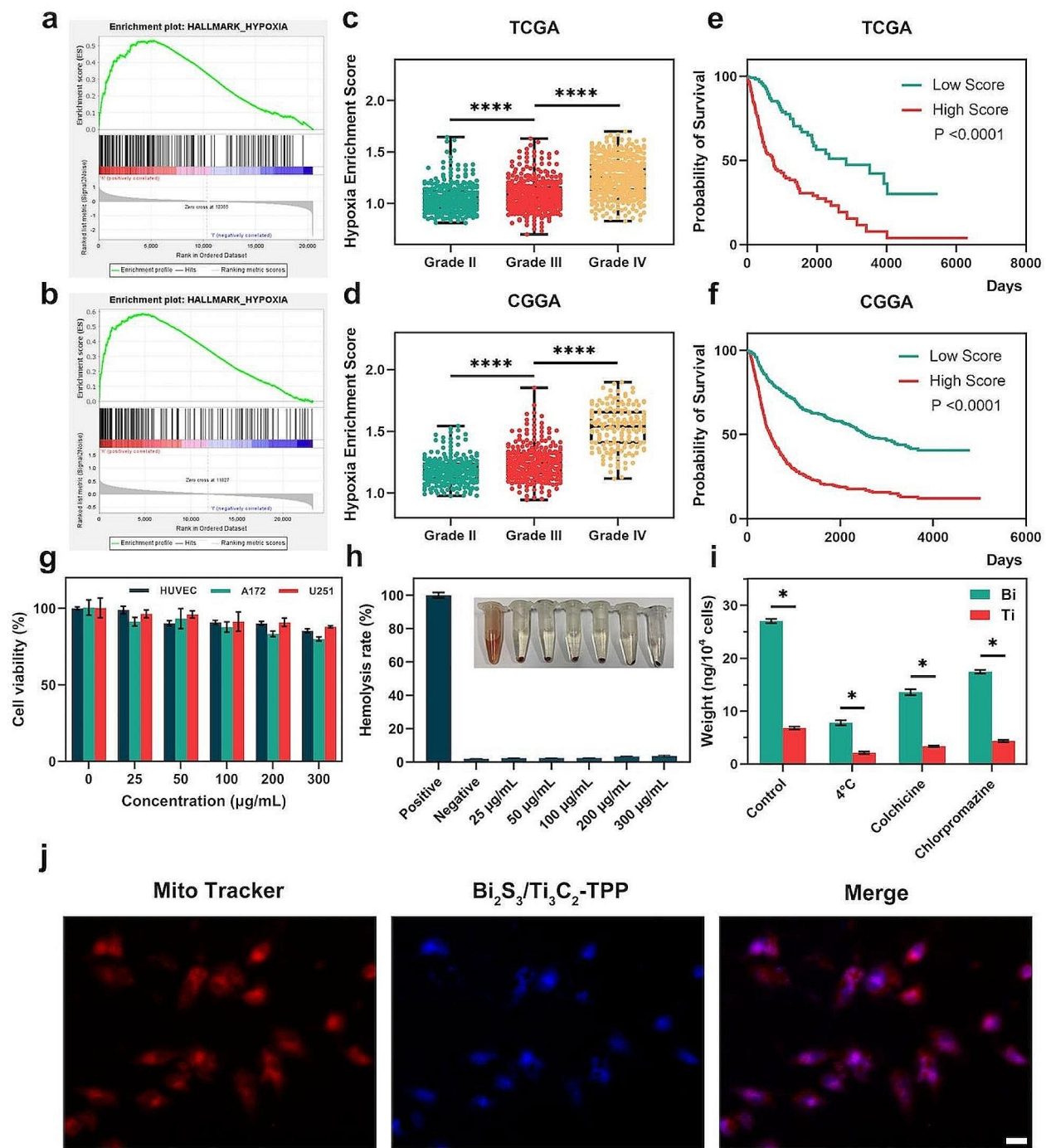


Fig. 3 Hypoxic tumor model selection, biocompatibility, and uptake of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$. The GSEA of hypoxia-related genes performed in **a** TCGA and **b** CGGA. The enrichment of hypoxia-related gene scores in gliomas of different grades in **c** TCGA and **d** CGGA; **** $P < 0.0001$. The relationship between hypoxia-related gene scores and patient survival time in **e** TCGA and **f** CGGA. **g** Cell viabilities of U251, A172, and HUVEC cells incubated with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ at various concentrations. **h** Hemolysis quantification of erythrocytes at various concentrations of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$. **i** The contents of Bi and Ti in U251 cells incubated with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ measured by ICP-MS; * $P < 0.05$. **j** The image of Mito-Tracker Red stained U251 cells treated with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ ($100 \mu\text{g mL}^{-1}$, scale bar = $20 \mu\text{m}$)

As a prerequisite for tumor phototherapy, effective cellular uptake is considered necessary to achieve a good therapeutic effect. To systematically investigate the pathways and conditions of the cellular internalization of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$, the content of two metallic elements, Bi and Ti, in tumor cells was determined by inductively coupled plasma-mass spectroscopy (ICP-MS) tests. As previously reported, low temperature could decrease adenosine triphosphate (ATP) synthesis and impede energy-dependent pathways [33]. As shown in Fig. 3i, when U251 cells were co-incubated with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ at 4°C , the content of Bi or Ti in the cells detected by ICP-MS tests was only about 30% of that under the incubation condition of 37°C , illustrating that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ endocytosis was primarily dependent on the energy-dependent pathways. Whereafter, colchicine and chlorpromazine (CPZ) were employed as macro-pinocytosis and clathrin-mediated endocytosis inhibitors to further explore the energy-dependent endocytosis mechanism of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$. The results indicated that, compared to the control group, colchicine- and CPZ-treated cells inhibited $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ uptake by 49.9% and 35.8% (Additional file 1: Fig. S11a), respectively, more definitely suggesting that the endocytosis pathway of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ was closely related to macro-pinocytosis and clathrin-mediated endocytosis. Moreover, after the incubation of U251 cells with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ or $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ for 4 h, the uptake of the two NMs in tumor cells showed a dose-dependent trend, with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ being more significantly taken up than $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ (Additional file 1: Fig. S11b, c). Thus, the targeting effect of TPP on tumor cells was confirmed. The tumor mitochondrial targeting performance of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ could also be examined by staining U251 cells with Mito-Tracker Red. It was clearly observed by an inverted fluorescence microscope that the inherent blue fluorescence of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ substantially overlapped with the red fluorescence of mitochondria (Fig. 3j), demonstrating that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ was principally localized in the mitochondria and had excellent mitochondrial targeting performance. Meanwhile, because of TPP's lipophilic cation characteristics, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ might also diffuse to other subcellular organelles, such as the endoplasmic reticulum and lysosomes [34, 35].

In vitro antitumor performance of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$

Mitochondrial depolarization could lead to the early apoptosis of cells [36]. JC-1 staining was used to monitor changes in mitochondrial membrane potential (MMP) to identify whether mitochondrial damage occurred following different experimental treatments. At normal MMP, JC-1 aggregated in the mitochondrial matrix, emitting red fluorescence, whereas at decreased MMP, it manifested a monomer state, emitting green fluorescence. As

shown in Fig. 4a, U251 cells in the phosphate buffered saline (PBS, pH=7.4, 10 mM), PBS with NIR irradiation (PBS+808 nm, 1 W cm^{-2} , 10 min), and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ groups mainly exhibited strong red fluorescence. Green fluorescence was found in the Bi_2S_3 , $\text{Bi}_2\text{S}_3+\text{Ti}_3\text{C}_2$, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ with NIR irradiation ($\text{Bi}_2\text{S}_3+808\text{ nm}$, $\text{Bi}_2\text{S}_3+\text{Ti}_3\text{C}_2+808\text{ nm}$, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2+808\text{ nm}$) groups, with the intensity of green fluorescence being stronger in the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2+808\text{ nm}$ group than in the first two groups. Furthermore, the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ with NIR irradiation ($\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}+808\text{ nm}$) group remarkably showed the strongest green fluorescence. The fluorescein isothiocyanate (FITC) channel of flow cytometry was performed to quantify the green fluorescence emitted by JC-1 monomer at decreased MMP in each group, and the results were consistent with the fluorescence images displayed above (Additional file 1: Fig. S12a). These studies illustrated that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ could effectively accumulate in mitochondria and induce severe mitochondrial dysfunction under NIR irradiation.

Because adequate O_2 supply could alleviate hypoxia in the TME and reduce resistance to the PDT effect, the O_2 -supplied capacity of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ in tumor cells during PDT was investigated. Image-iT[™], a red hypoxia probe, responded to decreased O_2 levels in living cells by increasing its red fluorescence signal intensity in real time, which was utilized to detect intracellular O_2 level. U251 cells in PBS with or without an 808 nm laser (1 W cm^{-2} , 10 min) irradiation exhibited similar red fluorescence intensity. However, cells treated by $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ with NIR irradiation exhibited notably lower red fluorescence intensity than those without NIR irradiation (Fig. 4b), demonstrating that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ had an excellent ability to split endogenous water and could produce sufficient O_2 under NIR irradiation. The intracellular O_2 supply of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ with NIR irradiation could further induce the generation of enough lethal ROS in tumor cells (Fig. 4c), so that it was able to achieve a favorable antitumor effect under hypoxia comparable to that under normoxia (Fig. 4d). Flow cytometry was conducted to quantify the intracellular ROS levels, employing DCF fluorescence. Compared to the PBS group, the changes in DCF fluorescence intensity detected in the PBS+808 nm and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ groups were negligible. The groups of $\text{Bi}_2\text{S}_3+808\text{ nm}$, $\text{Bi}_2\text{S}_3+\text{Ti}_3\text{C}_2+808\text{ nm}$, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2+808\text{ nm}$ could elicit DCF fluorescence enhancement, while the first two groups only displayed modest enhancement. Nevertheless, the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}+808\text{ nm}$ group considerably improved the DCF fluorescence intensity (Additional file 1: Fig. S12b), confirming the finding that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ could generate abundant ROS by offering O_2 under NIR irradiation to cause more serious oxidative injury to tumor cells.

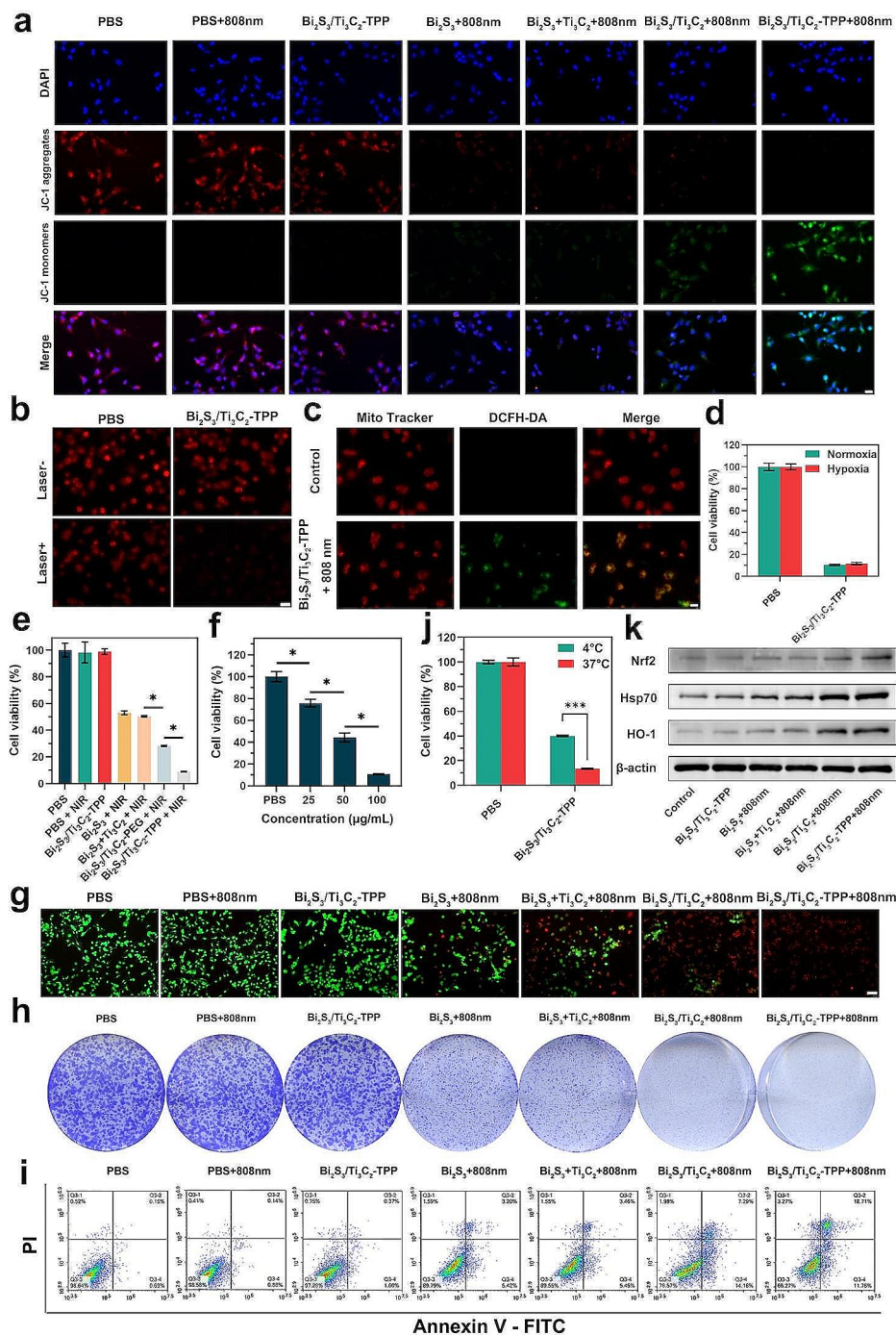


Fig. 4 In vitro antitumor performance evaluation of Bi₂S₃/Ti₃C₂-TPP. **a** The images of JC-1 stained U251 cells after different treatments (scale bar = 20 μm). **b** Hypoxia level of PBS (pH=7.4, 10 mM) and Bi₂S₃/Ti₃C₂-TPP (100 μg mL⁻¹) treated U251 cells (scale bar = 50 μm). **c** ROS generation in U251 cells under 808 nm laser irradiation (1 W cm⁻², 10 min, scale bar = 20 μm). **d** U251 cells treated with PBS (pH=7.4, 10 mM) or Bi₂S₃/Ti₃C₂-TPP (100 μg mL⁻¹) under 808 nm laser irradiation (1 W cm⁻², 10 min) in N₂ or air atmospheres. **e** MTT and **g** Calcein AM/PI two-staining analysis of U251 cells following various treatments (scale bar = 100 μm); *P<0.05. **f** The viabilities of U251 cells treated with Bi₂S₃/Ti₃C₂-TPP at various concentrations under 808 nm laser irradiation (1 W cm⁻², 10 min); *P<0.05. **h** Cell clone formation of U251 cells after different treatments. **i** Flow cytometric analysis for apoptotic U251 cells by Annexin-V-FITC/PI assay. **j** U251 cells treated with PBS (pH=7.4, 10 mM) or Bi₂S₃/Ti₃C₂-TPP (100 μg mL⁻¹) under 808 nm laser irradiation (1 W cm⁻², 10 min) at 37 °C and 4 °C; ***P<0.001. **k** Cellular Nrf2, Hsp70, and HO-1 expressions of U251 cells following various treatments

The antitumor properties of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ under NIR irradiation in vitro were thoroughly researched. According to MTT assays (Fig. 4e), the antitumor effects of the PBS, PBS+808 nm, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ groups were marginal. The tumor cell viabilities of the Bi_2S_3 +808 nm, $\text{Bi}_2\text{S}_3+\text{Ti}_3\text{C}_2$ +808 nm, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ +808 nm groups were 52.9%, 50.5%, and 28.2%, respectively. Moreover, the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ +808 nm group showed the best antitumor effect, with tumor cell viability reduced to 8.9%. The tumor cell viabilities of different concentrations of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ under 808 nm laser (1 W cm^{-2} , 10 min) irradiation were also examined (Fig. 4f), suggesting that $100 \mu\text{g mL}^{-1}$ was a proper concentration for antitumor therapy. Live/dead cell staining analysis further intuitively revealed that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ under NIR irradiation could lead to more significant tumor cell death, which was compatible with the MTT assay results (Fig. 4g). Besides, the clone formation assay could reflect cell proliferation ability, which was carried out to discover that the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ +808 nm group possessed a lower cloning density and a substantially smaller number of cloned clusters than the control group (Fig. 4h), implying that this cell-treated condition had a strong inhibitory effect on tumor cell clone formation. Apoptosis was also found to be an essential marker of oxidative stress injury. Thus, flow cytometry was used to illustrate that there were almost no apoptotic and necrotic tumor cells in the PBS, PBS+808 nm, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ groups (Fig. 4i). The proportions of apoptosis and necrosis triggered by Bi_2S_3 +808 nm, $\text{Bi}_2\text{S}_3+\text{Ti}_3\text{C}_2$ +808 nm, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ +808 nm groups were 10.21%, 10.45%, and 23.43%, respectively, with the highest being 33.73% in the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ +808 nm group. Likewise, these outcomes confirmed that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ under NIR irradiation could inhibit tumor cell proliferation by promoting their apoptosis, showing an outstanding in vitro antitumor effect. To determine the importance of ROS in the phototherapeutic effect of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$, the contribution of PDT to total tumor cell death was assessed by performing MTT assays at 4°C to restrain the PTT effect. As shown in Fig. 4j, the injury of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ under NIR irradiation to tumor cells was weakened at 4°C compared to 37°C , and the percentage of tumor cell death induced by PDT accounted for approximately 69.2%, emphasizing the leading role of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ photodynamic performance in the antitumor process. All of these findings validated that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ had fantastic in vitro antitumor capability, which was achieved through the synergistic and efficient therapeutic effect of multimode PDT combined with PTT.

Aiming to clarify the mechanism by which $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ caused tumor cell injury, western blot analysis was conducted on U251 cells treated under different experimental conditions to explore the relevant

protein expressions. Nuclear factor erythroid 2-related factor 2 (Nrf2) is a basic leucine zipper protein that regulates the expression of antioxidant proteins like heme oxygenase-1 (HO-1), which protect organisms from oxidative damage via an antioxidant reaction. These two proteins' expression could reflect the cellular damage induced by PDT. The expression of heat-responsive heat shock protein 70 (Hsp70) was frequently utilized to evaluate the damage caused by PTT-triggered intracellular hyperthermia. As shown in Fig. 4k and Additional file 1: Fig. S13, the expressions of Nrf2, HO-1, and Hsp70 in the group of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ without an 808 nm laser (1 W cm^{-2} , 10 min) irradiation were not significantly different from those in the control group. The expressions of these three proteins in the Bi_2S_3 +808 nm, $\text{Bi}_2\text{S}_3+\text{Ti}_3\text{C}_2$ +808 nm, and $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ +808 nm groups were all elevated compared to the control group, and the expressions in the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ +808 nm group were higher than those in the first two groups. Additionally, the expressions of Nrf2, HO-1, and Hsp70 in the $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ +808 nm group were the highest. The results indicated that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ could more effectively stimulate the oxidative stress pathway and had a superior thermal response under NIR irradiation.

Biosafety and biodistribution of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$

Prior to exploring the in vivo antitumor properties of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$, we investigated its potential toxicity to better understand the biosafety of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$. Routine blood biochemical indices of mice were detected, including alanine aminotransferase (ALT), alanine aminotransferase (AST), albumin (ALB), globulin (GLOB), total protein (TP), blood urea nitrogen (BUN), and creatinine (CREA), and the ratio of ALB to GLOB (A/G) was calculated. As shown in Fig. 5a, the mice treated with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ were not abnormal compared to the control group, suggesting that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ had no evident hepatic and renal cytotoxicity. Furthermore, the major organs (heart, liver, spleen, lung, and kidney) of each group after 2 weeks of experimental treatments were sliced, and hematoxylin and eosin (H&E) staining was performed. The results (Fig. 5b) demonstrated no substantial inflammation, damage, or chronic pathological toxicity, verifying the good biocompatibility of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ in vivo. Both studies showed that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ had favorable biosafety.

The blood circulation of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ was consistent with a two-compartment model, and its half-life was calculated to be approximately 4.3 h (Fig. 5c), allowing $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ to achieve efficient accumulation at the tumor site for treatment. Similarly, the contents of Bi and Ti elements in major organs and tumors of mice collected after caudal vein injection of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ for 24 h were detected by ICP-MS tests (Fig. 5d). High

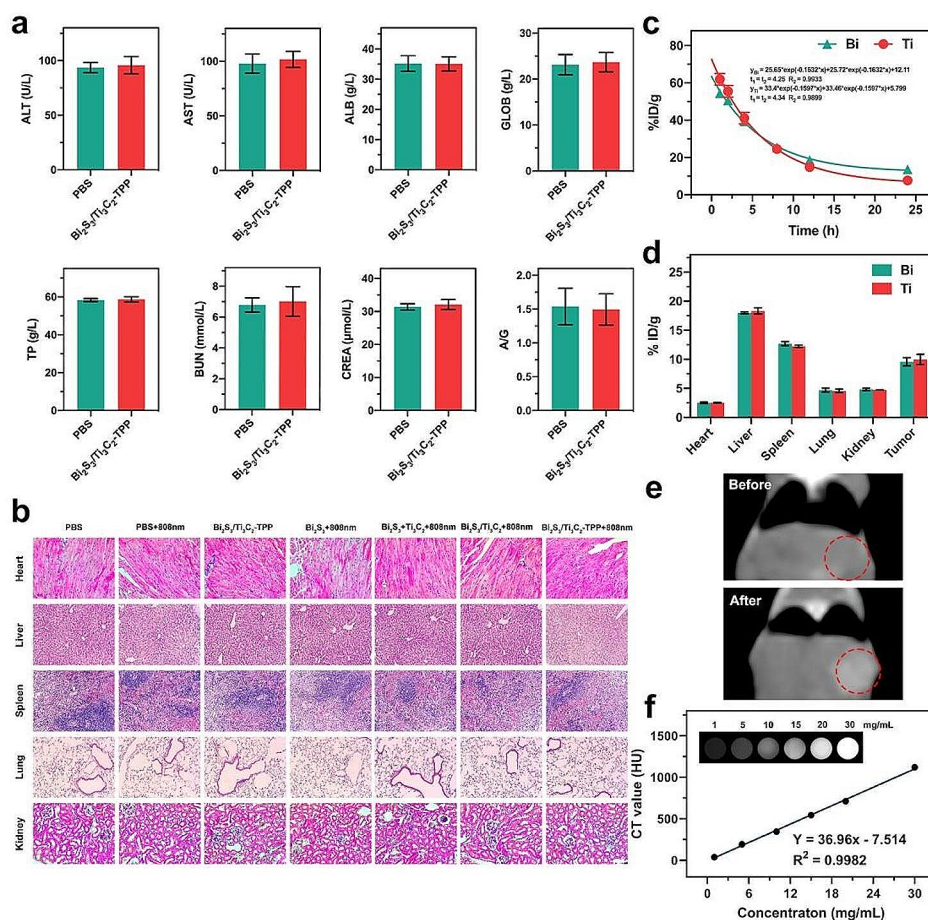


Fig. 5 Biosafety and biodistribution evaluation of Bi₂S₃/Ti₃C₂-TPP. **a** Serum biochemistry results involving ALB, ALT, AST, TP, GLOB, BUN, CREA, and A/G ratio obtained from mice injected with PBS (pH=7.4, 10 mM) or Bi₂S₃/Ti₃C₂-TPP (20 mg kg⁻¹) at 24 h. **b** The pictures of H&E dyed primary apparatus of mice involving heart, liver, spleen, lung, and kidney following various treatments (scale bar = 40 μm). **c** Blood circulation and **d** biodistribution of Bi₂S₃/Ti₃C₂-TPP (20 mg kg⁻¹) after intravenous injection into U251 tumor-bearing mice by measuring Bi and Ti concentrations with ICP-MS. **e** In vivo CT pictures of U251 tumor-bearing mice before and after injecting with Bi₂S₃/Ti₃C₂-TPP (20 mg kg⁻¹) intravenously. **f** In vitro CT images and CT values of Bi₂S₃/Ti₃C₂-TPP at various concentrations

tumor accumulation of Bi₂S₃/Ti₃C₂-TPP (~ 9.78% ID/g) was found due to the enhanced permeability and retention (EPR) effect, and the high levels of it in the major metabolic organ, liver and spleen, revealed that Bi₂S₃/Ti₃C₂-TPP could be cleared from the organism through these two organs. CT images of mice at 8 h post-injection further confirmed the significant accumulation of Bi₂S₃/Ti₃C₂-TPP in tumor tissues (Fig. 5e, f), indicating that Bi₂S₃/Ti₃C₂-TPP could be employed as a CT imaging agent to benefit in tumor diagnosis and guided treatment, ensuring a satisfactory tumor therapeutic outcome.

In vivo antitumor Performance of Bi₂S₃/Ti₃C₂-TPP

Bi₂S₃/Ti₃C₂-TPP showed excellent synergistic effect of PDT and PTT in tumor cells in vitro, implying that Bi₂S₃/Ti₃C₂-TPP would have an outstanding tumor therapeutic effect in vivo. Balb/c U251 tumor-bearing nude mice were randomly divided into 7 treatment groups

(*n*=5) for different treatments to investigate the antitumor performance of Bi₂S₃/Ti₃C₂-TPP in vivo: Group I: caudal vein injection of PBS (pH=7.4, 10 mM); Group II: PBS with an 808 nm laser irradiation (1 W cm⁻², 10 min); Group III: Bi₂S₃/Ti₃C₂-TPP (20 mg kg⁻¹); Group IV: Bi₂S₃ with an 808 nm laser irradiation; Group V: Bi₂S₃+Ti₃C₂ with an 808 nm laser irradiation (dose ratio 3:2); Group VI: Bi₂S₃/Ti₃C₂ with an 808 nm laser irradiation; Group VII: Bi₂S₃/Ti₃C₂-TPP with an 808 nm laser irradiation. Firstly, the O₂ production capacity of Bi₂S₃/Ti₃C₂-TPP in vivo under NIR irradiation was examined by detecting the expression of hypoxia-inducible factor-1α (HIF-1α) in tumor tissues. HIF-1α expression in the tumor tissues of Group VII was dramatically down-regulated, suggesting that the tumor hypoxic microenvironment was successfully alleviated (Fig. 6a and Additional file 1: Fig. S14). Subsequently, temperatures of the tumor sites in mice under NIR irradiation were measured with a portable

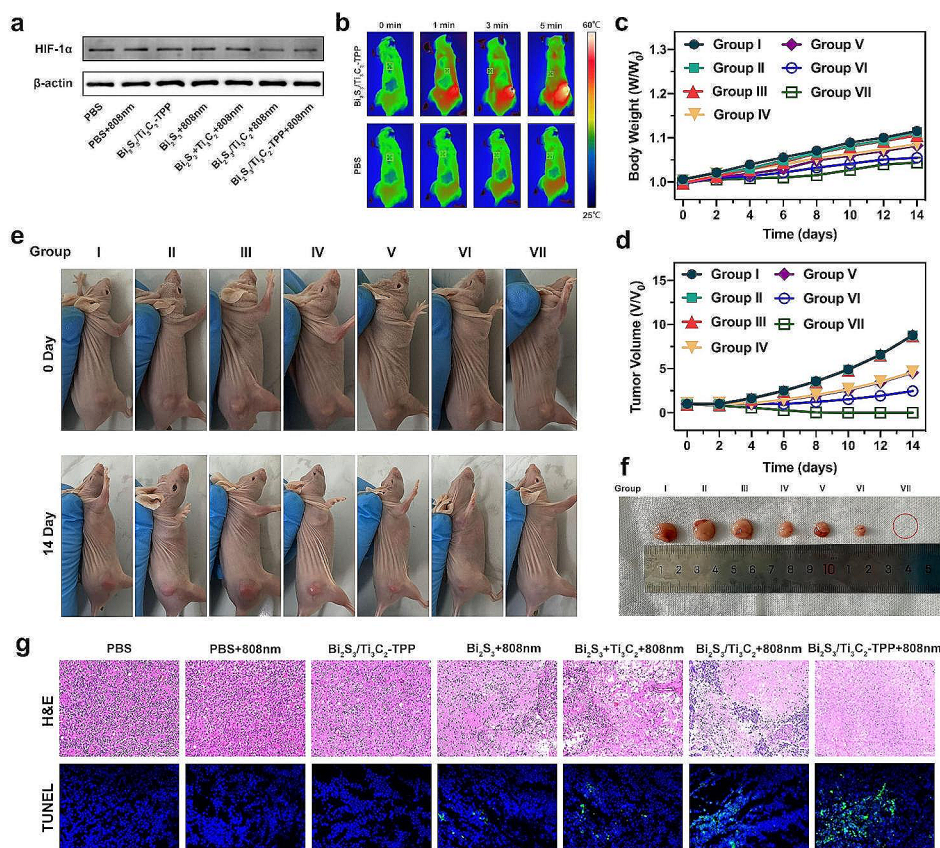


Fig. 6 In vivo antitumor performance evaluation of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$. **a** HIF-1 α expressions of tumors after different treatments. **b** Infrared thermography of U251 tumor-bearing mice after being injected with PBS (pH = 7.4, 10 mM) or $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ (20 mg kg^{-1}). Time-dependent relative **c** body mass and **d** tumor dimension change plots from various groups. **e** The photos of U251 tumor-bearing mice from various groups during two weeks. **f** The representative photo of tumors dissected at 14 days post-injection. **g** Representative images of H&E (scale bar = 100 μm) and TUNEL (scale bar = 20 μm) stained slices of tumor tissues after different treatments

thermal imager and recorded, aiming to explore the photothermal performance in vivo. As shown in Fig. 6b and Additional file 1: Fig. S15, the tumor-site temperature of mice treated with $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ increased from 31.2 $^\circ\text{C}$ to 51.9 $^\circ\text{C}$ after an 808 nm laser irradiation (1 W cm^{-2}) for 5 min, demonstrating that $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ had strong photothermal conversion ability in vivo. Meanwhile, the body weights and tumor volumes of the mice were monitored every other day within two weeks after treatments, and photographs were taken to evaluate the therapeutic effect of each group. According to the results (Fig. 6c-f), the control groups (Group I, II, and III) had no apparent inhibitory effect on tumor growth in mice. Groups IV, V, and VI were observed to have a therapeutic impact on mouse tumors, and the inhibition degree of tumor growth in Group VI was higher than in Group IV and V owing to the combination of multimodal PDT and PTT. Notably, the tumor growth in group VII was entirely inhibited, attributed to the tumor mitochondria-targeting property conferred by TPP, and no tumor recurrence was found over two weeks, which illustrated

the exceptional phototherapy effect of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ under NIR irradiation. Furthermore, there was no noteworthy change in mouse body weight amongst all experimental groups during two weeks, indicating that the systemic side effects of NMs were negligible. H&E and transferase-mediated deoxyuridine triphosphate nick end labeling (TUNEL) staining results (Fig. 6g) revealed that no inflammation or damage was observed in tumor cells in the control groups (Group I, II, and III). There was tiny focal or fragmentary tumor cell necrosis in Group IV, V, and VI, and the necrosis degree in Group VI was more severe than in Group IV and V, while Group VII had the largest proportion of tumor cell necrosis. These further confirmed that the in vivo anti-tumor performance of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ was extraordinary under NIR irradiation.

Conclusion

In summary, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2\text{-TPP}$ was designed to produce O_2 and ROS synchronously via type I and type II PDT mechanisms, combined with PTT under NIR irradiation

at the same wavelength for synergistic phototherapy of hypoxic tumors. The formation of $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs not only enhanced the separation efficiency of photoinduced electrons and holes, but also endowed them with strong redox capacity. Actually, after the modification of TPP on $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ NHs, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP showed favorable biocompatibility and could be effectively internalized into tumor cells by targeting the mitochondria, which induced more severe tumor cell damage and elimination under NIR irradiation and exhibited excellent CT imaging capability. Therefore, $\text{Bi}_2\text{S}_3/\text{Ti}_3\text{C}_2$ -TPP expands the phototherapeutic effect of Bi_2S_3 -based nanoplateforms and holds a great prospect in hypoxic tumor theranostics.

Supplementary Information

The online version contains supplementary material available at <https://doi.org/10.1186/s12951-024-02391-x>.

Supplementary Material 1

Author contributions

HJ and JS wrote the main manuscript and completed the main analysis work. FL, YZ, XC, and CD participated in the design of this study and funded the study methods. ZW were mainly in charge of validation, supervision, and funding acquisition. All authors reviewed the manuscript.

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Data availability

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Declarations

Competing interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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