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Simvastatin induced ferroptosis for triple-negative breast cancer therapy

Xianxian Yao¹, Ruihong Xie¹, Yongbin Cao¹, Jing Tang², Yongzhi Men³, Haibao Peng^{4*} and Wuli Yang^{1*}

Abstract

Triple-negative breast cancer (TNBC), a management of aggressive breast cancer, remains an unmet medical challenge. Although a wave of efforts had spurred to design novel therapeutic method of TNBC, unpredictable prognosis with lacking effective therapeutic targets along with the resistance to apoptosis seriously limited survival benefits. Ferroptosis is a non-apoptotic form of cell death that is induced by excessive lipid peroxidation, which provide an innovative way to combat cancer. Emerging evidence suggests that ferroptosis plays an important role in the treatment of TNBC cells. Herein, a novel ferroptosis nanomedicine was prepared by loading simvastatin (SIM), a ferroptosis drug, into zwitterionic polymer coated magnetic nanoparticles (Fe₃O₄@PCBMA) to improve the therapeutic effect of TNBC. The as-obtained Fe₃O₄@PCBMA-SIM nanoparticles demonstrated more cytotoxicity against MDA-MB-231 than MCF-7 due to the higher expression of 3-hydroxy-3-methyl-glutaryl-coenzyme A reductase (HMGCR), which demonstrated that statins could effectively kill TNBC. Further experiments showed that SIM could inhibit the expression of HMGCR to downregulate the mevalonate (MVA) pathway and glutathione peroxidase 4 (GPX4), thereby inducing cancer cell ferroptosis. What's more, PCBMA endows Fe₃O₄@PCBMA longer blood circulation performance to enhance their accumulation at tumor sites. Given that Fe₃O₄ have proven for clinical applications by the U.S. Food and Drug Administration (FDA) and SIM could induce cancer cell ferroptosis, the developed Fe₃O₄@PCBMA-SIM nanosystem would have great potential in clinics for overcoming the drug resistance brought about by apoptotic drugs to cancer cells.

Keywords: Triple-negative breast cancer, Ferroptosis, Simvastatin, Long circulation, Controlled release

Introduction

Breast cancer has the highest fatality rate in women worldwide [1]. Among all types and forms of breast cancer, triple negative breast cancer (TNBC) is the most aggressive and heterogeneous subtype due to lack of estrogen and progesterone receptor expressions, which has been an unmet medical challenge in clinic [2, 3]. The distinct cellular phenotype with lack of receptor or target makes chemotherapy as an excellent treatment for

TNBC. Nonetheless, the drug resistance and toxic side effects resulting from anti-cancer drugs lead to the failure of cancer chemotherapy, which due to the acquired or intrinsic resistance of cancer cells to apoptosis [4]. Hence, develop effective non-apoptotic treatment strategies has become an urgent need for the treatment of TNBC.

Ferroptosis is a form of iron-dependent cell death induced by excessive lipid peroxidation that distinct from the traditional apoptosis and necrosis [5]. Since the term ferroptosis was proposed in 2012, the unique mechanism of ferroptosis has attracted increasing attention in the field of antitumor therapy [6]. The redoxactive iron (Fe²⁺) is the key elements of ferroptosis and this process characterized by direct or indirect inhibition of glutathione peroxidase 4 (GPX4), lipid repair enzyme, and lipid hydroperoxides (LPO) accumulation

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[7, 8]. Consequently, the intracellular accumulation of LPO leading to impaired cell structure and integrity [9]. The inactivation of GPX4 was produced in the presence of erastin analogs or the direct GPX4 inhibitor [10], such as RSL-3 [11], sorafenib [9], statins [12] and so on. Erastin is a low molecular chemotherapeutics agent and TNBC cells have been reported to be sensitive to erastin-induced ferroptosis using xCT cystine/glutamate antiporter as a common therapeutic target for TNBC [13, 14]. To date, several mechanisms leading to iron and reactive oxygen species (ROS) metabolism of ferroptosis have been addressed, but the mechanisms of ferroptosis in breast cancer cells especially in TNBC has hardly been reported [15]. Moreover, it is still challenging to engineer the iron ion delivery system to enhancement the effect of ferroptosis. Therefore, improving the efficiency of iron ion delivery in the process of ferroptosis is of great significance to TNBC patient for cancer treatment and drug design [10].

Statins are a class of low molecular weight drugs that have been approved for clinical control of hypercholesterolemia [16]. Recent study has shown that statins have a potential role in cancer prevention due to their ability to inhibit proliferation, angiogenesis and inflammation [12, 17]. Joseph et al. reported that statins could lower cholesterol through inhibit 3-hydroxy-3-methyl-glutarylcoenzyme A reductase (HMGCR) to regulate the mevalonate (MVA) pathway [18]. Moreover, it was reported that isopentenyl pyrophosphate participant in the biosynthesis of GPX4 through MVA pathway and HMGCR played a vital role in the synthesis of isopentenyl pyrophosphate [19, 20]. So far, ferroptosis has been thought to induce cancer cell oxidative damage by controlling the phospholipid hydroperoxide-reducing enzyme GPX4 [19, 21, 22]. Therefore, it will be of great significance if statins could kill TNBC through the way of ferroptosis. However, statins as a kind of small molecule drug is metabolized quickly and few drugs accumulate to the lesion site, severely reducing the effectiveness of treatment. Nanoscale sized materials as an excellent carrier could deliver antitumor drugs to tumor tissues through passively target [23]. But most of nanoparticles rapidly cleared by the reticuloendothelial system as exogenous invaders that affect the percentage of administered nanoparticles reaching in tumor site and limited therapeutic effect [24, 25]. Therefore, development of novel nanomedicine with long circulation that can enhance the statins accumulation for in vivo TNBC therapy is urgently required.

In this work, we present the construction of zwitterionic polymer coating of ferroferric oxide nanoparticles (Fe₃O₄@PCBMA) to prolong its blood circulation. Simvastatin (SIM), a ferroptosis drug, could be loaded into

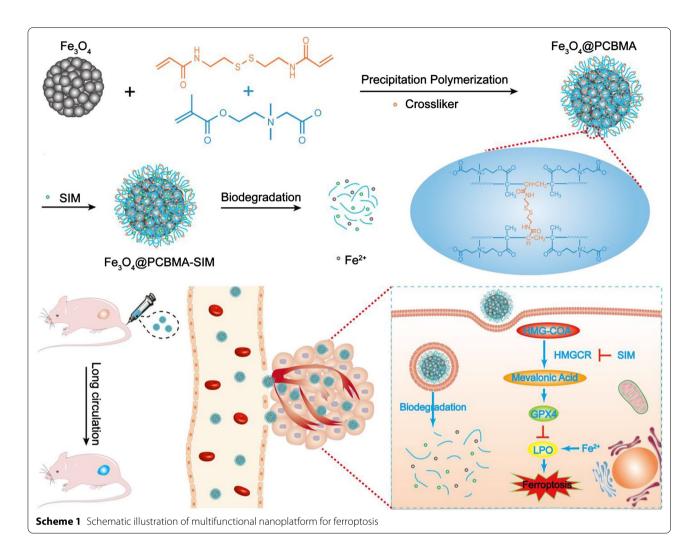
 ${\rm Fe_3O_4@PCBMA}$ (${\rm Fe_3O_4@PCBMA\text{-}SIM}$). MDA-MB-231, a TNBC model, and MCF-7, a normal breast cancer cell model, were used to evaluate the cancer cell killing efficiency. The results showed SIM have more cytotoxicity against MDA-MB-231 than MCF-7, which demonstrated that statins could effectively kill TNBC. In addition, the western blot result illustrated SIM could through inhibit HMGCR to modulate the MVA pathway and deactivate GPX4. With the inherited blood circulation property and ferroptosis effect, the in vivo therapeutic efficiency of ${\rm Fe_3O_4@PCBMA\text{-}SIM}$ was evaluated through building MDA-MB-231 tumor-bearing mice. Our finding highlights that ${\rm Fe_3O_4@PCBMA\text{-}SIM}$ exhibit an excellent tumor suppression, which will open an avenue of TNBC therapy.

Results and discussion

The preparation of Fe₃O₄@PCBMA

Carboxybetaine methacrylate (CBMA) was synthesized according to the reported method [26]. Then, we encapsulated Fe₃O₄ with poly (carboxybetaine methacrylate) (PCBMA) for enabling longer blood circulation performance [27]. The preparation process of Fe₃O₄@PCBMA was illustrated in Scheme 1. Before encapsulation, carbon-carbon double bond was firstly modified to magnetic nanoparticle, which was proved at the strong peaks centered of 1717 cm⁻¹ by fourier transform infrared spectroscopy (FTIR) (Additional file 1: Fig. S1). Then, 3-aminopropyltriethoxysilanemodified Fe₃O₄ (Fe₃O₄-MPS) nanoparticles were coated with PCBMA network by reflux precipitation polymerization method [28]. Transmission electron microscope (TEM) images of Fe₃O₄ and Fe₃O₄@PCBMA with uniformed size displayed spherical morphology and the obvious polymer layer (Fig. 1A, B). In addition, the smooth surface of nanoparticles after coating from the scanning electron microscope also indicated the form of core-shell structure (Fig. 1C, D). It could be seen from the TEM image that the thickness of the zwitterionic polymer layer was about 8 nm. Meanwhile, the hydrate particle size of Fe₃O₄@PCBMA was a little bigger than Fe₃O₄ measured by dynamic light scattering (DLS), which was further prove the successful coating (Fig. 1E). In addition, there was no different of DLS particle size that Fe₃O₄@PCBMA dispersed in aqueous, phosphate buffer solution (PBS), bull serum albumin (BSA) and culture medium Dulbecco Minimum Essential Medium (DMEM), which indicated Fe₃O₄@ PCBMA have the good stability (Additional file 1: Fig. S3). Furthermore, the zeta potential of Fe₃O₄@PCBMA increased to zero after coating, which due to the equal positive and negative of PCBMA (Fig. 1F). Therefore,

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the above results indicated the successful fabrication of Fe₃O₄@PCBMA nanoparticles. And then the SIM loading property was studied. After SIM loading, the morphology and size of the nanoparticles barely changed, indicating the stable of the drug-loaded nanoparticles (Additional file 1: Fig. S2). The FTIR spectra of the Fe₃O₄-SIM and Fe₃O₄@PCBM-SIM appeared new bands in the 1300-1000 cm⁻¹ region corresponding to C-O-C of SIM. In addition, there are 11.4% weight loss of PCBMA at 200-400 °C and 15.7% weight loss of SIM at 200–350 °C showed by thermogravimetric analysis (TGA) of Fe₃O₄@PCBMA-SIM. Moreover, Fig. 1I showed the UV-vis absorbance spectra of the nanoparticles before and after loading SIM. The absorbance peak of Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM at 238 nm were attributed to the SIM characteristic absorbance, which indicated the SIM successful loading. These results suggested the successful fabrication of Fe₃O₄@ PCBM-SIM through PCBMA coating and SIM loading.

The biodegradation and controlled drug release

A sustainable and efficient fenton reaction nanoplatform for tumor therapy was developed based on Fe₃O₄ nanoparticle. As described in Fig. 2A, Fe₃O₄@PCBMA-SIM nanoparticles could trigger more ROS generation in response to tumor microenvironment and release large amounts of Fe²⁺ for further promoting cancer cell death. In addition, the concentration of glutathione (GSH) in cancer cell was ranging from 2 to 10 mM and the pH was 6.5-5.0 in lysosomes of cancer cell and 7.4 in normal tissues [29, 30]. Therefore, the biodegradability of Fe₃O₄ and Fe₃O₄@PCBMA were investigated. Certified by the inductively coupled plasma spectrometry (ICP-AES), the degradation property of Fe₃O₄@PCBMA were increased by the increased concentrations of GSH and as the pH value of PBS decreased. After 96 h, the Fe concentration was only 1.8 µg/mL in the pH 5.0 buffer solution with GSH of 10 mM and there was only few Fe in neutral environment with GSH of 10 mM, implying that Fe₃O₄@ PCBMA could be decomposed into iron ions (Fig. 2B).

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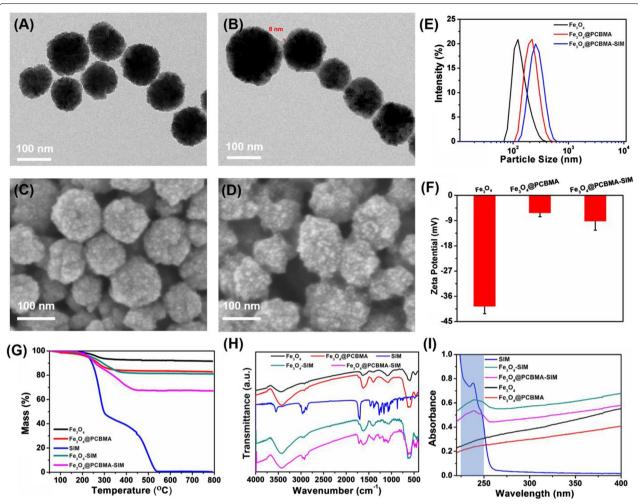


Fig. 1 TEM images of Fe_3O_4 (**A**) and Fe_3O_4 @PCBMA (**B**) nanoparticles; SEM images of Fe_3O_4 (**C**) and Fe_3O_4 @PCBMA nanoparticles (**D**); DLS diameters of Fe_3O_4 . Fe_3O_4 @PCBMA, and Fe_3O_4 @PCBMA-SIM nanoparticles (**E**); Zeta potential of Fe_3O_4 . Fe_3O_4 @PCBMA, and Fe_3O_4 @PCBMA-SIM nanoparticles (**F**); TGA analysis of Fe_3O_4 . Fe_3O_4 @PCBMA, SIM, Fe_3O_4 -SIM and Fe_3O_4 . Fe_3O_4 PCBMA-SIM nanoparticles measured in air (**G**); FT-IR spectra of Fe_3O_4 . Fe_3O_4 PCBMA, SIM, Fe_3O_4 -SIM and Fe_3O_4 @PCBMA-SIM, Fe_3O_4 =SIM, Fe_3O_4 =SIM, Fe_3O_4 =CBMA-SIM, Fe_3O_4 =CBMA nanoparticles (**I**):

The degradation property of Fe $_3$ O $_4$ was similar to Fe $_3$ O $_4$ @ PCBMA and the Fe concentration (2.3 µg/mL) in the pH 5.0 buffer solution with GSH of 10 mM was higher than Fe $_3$ O $_4$ @PCBMA, which due to the existence of PCBMA in the shell (Additional file 1: Fig. S4). Therefore, SIM would be released faster after phagocytosis by cancer cells in theory.

To evaluate drug release behavior, we first measured the drug loading ratios of SIM by UV–vis absorbance spectra at 238 nm. The UV–vis calculated that SIM was loaded in Fe₃O₄ and Fe₃O₄@PCBMA with contents of 10 and 15% respectively. Then the mass ratio of nanoparticles to SIM was deliberately controlled to 10% for the consistency of subsequent experiments. Afterwards, the drug release property of Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM

nanoparticles were studied by dispersing the nanoparticles into glutathione (GSH, 10 mM). As shown in Fig. 2C, the SIM releasing amount increased with incubating time and there were about 70% SIM released from Fe $_3$ O $_4$ -SIM and 55% SIM released from Fe $_3$ O $_4$ @PCBMA-SIM over 24 h, indicating a distinct rapid release behavior. Therefore, Fe $_3$ O $_4$ -SIM and Fe $_3$ O $_4$ @PCBMA-SIM could release SIM under the microenvironments of cancer cell.

In-vitro cytotoxicity and cell uptake of Fe₃O₄@PCBMA-SIM

It is well known that TNBC is very difficult to treatment owing to its heterogeneity, molecular variability, and stemness [31]. It is very significant to develop sufficient drug to kill TNBC. The viability of breast cancer cell (MCF-7) and triple-negative breast cancer cell

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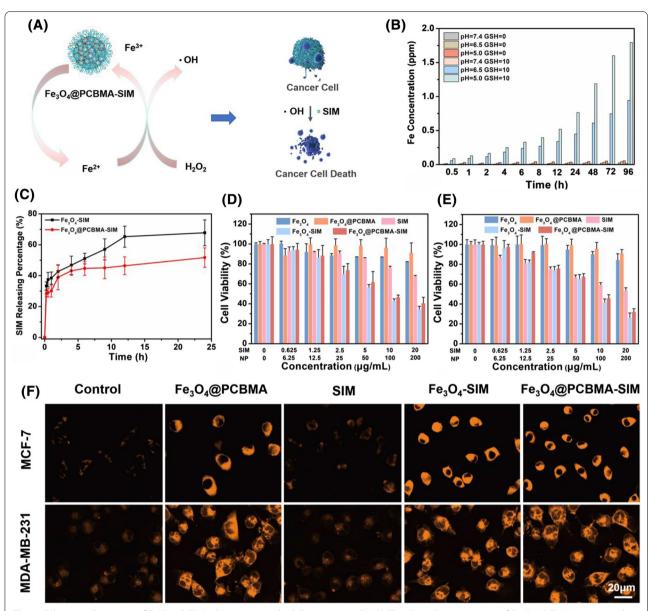


Fig. 2 Schematic illustration of Fe₃O₄@PCBMA-SIM nanoparticles killing cancer cells (**A**); The degradation property of Fe₃O₄@PCBMA nanoparticles dispersed in different concentrations of GSH (0 mM and 10 mM) and pH values (5.0, 6.5 and 7.4) (**B**); Cumulative SIM releasing from Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM nanoparticles dispersed in 10 mM GSH (**C**); Cell viability of the MCF-7 (**D**) and MDA-MB-231 (**E**) cells after 48 h incubation with Fe₃O₄, Fe₃O₄@PCBMA, SIM, Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM suspensions; CLSM of control, Fe₃O₄@PCBMA, free SIM, Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM nanoparticles after uptake by MCF-7 and MDA-MB-231 cells (**F**)

(MDA-MB-231) treated with Fe $_3$ O $_4$ @PCBMA-SIM were investigated in this study adopting the cell counting kit-8 (CCK-8) assay. Before in vitro cytotoxicity experience, we first studied the biocompatibility. As shown in Fig. 2D, E, the viability of two cancer cells cultured in the presence of Fe $_3$ O $_4$ were indicated slightly toxicity. After encapsulate PCBMA, there were negligible cytotoxicity of Fe $_3$ O $_4$ @PCBMA even though the concentration of nanoparticles were 200 μ g/mL. In addition, free SIM, Fe $_3$ O $_4$ -SIM

and Fe $_3$ O $_4$ @PCBMA-SIM displayed a concentration-dependent cytotoxicity against both MCF-7 and MDA-MB-231 cells, and free SIM, Fe $_3$ O $_4$ -SIM and Fe $_3$ O $_4$ @PCBMA-SIM were more cytotoxic to MDA-MB-231 cells than to MCF-7 cells, which proved that Fe $_3$ O $_4$ @PCBMA-SIM could effectively kill TNBC. For example, the cell viability of MDA-MB-231 was 54% after treated with free SIM (20 μ g/mL), which was lower than that MCF-7 cells (68%). The cytotoxicity to MDA-MB-231 (32%) still

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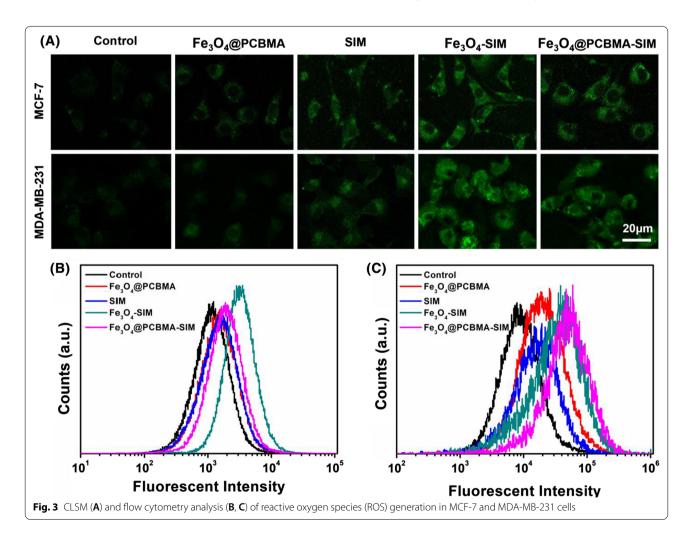
higher than to MCF-7 cells (41%) after loaded to Fe₃O₄, probably due to the higher contribution of SIM and Fe₃O₄ to the system than free SIM. However, Fe₃O₄-SIM exhibited slightly higher cytotoxicity to two cancer cells than Fe₃O₄@PCBMA-SIM, which was due to the shell protection of PCBMA. In addition, the cell viability of MCF-7 cells and MDA-MB-231 cells with ferrous sulfate were measured. The cell viability was no obvious difference after ferrous sulfate added compared to control group whether ferrous sulfate added to MCF-7 cells or MDA-MB-231 cells, which confirmed that the iron ion has no cytotoxicity against MDA-MB-231 (Additional file 1: Fig. S5) and MCF-7 (Additional file 1: Fig. S6). Therefore, it can be concluded that SIM was higher sensitive to MDA-MB-231 and Fe₃O₄@PCBMA-SIM was an excellent strategy for the treatment of TNBC.

Further, the phagocytosis property of nanoparticles was investigated using FerroOrange as iron ion detection reagent in cells [32]. As shown in Fig. 2F, the orange fluorescence were obvious when MCF-7 and MDA-MB-231

cells treated with Fe $_3$ O $_4$, Fe $_3$ O $_4$ -SIM and Fe $_3$ O $_4$ @PCBMA-SIM nanoparticles, whereas few orange fluorescence in control group and free SIM group, indicating that enormous amount of Fe $^{2+}$ ions generated via nanoparticles. Therefore, Fe $_3$ O $_4$ @PCBMA-SIM nanoparticles could be degraded into iron ion after phagocytosed by cancer cells.

In-vitro reactive oxygen species (ROS) generation

It was well-known that ROS could generated by ferrous ions via fenton reaction. To visually observe the generation of ROS, dichlorofluorescein diacetate (DCFH-DA) was used as a fluorescent probe to detect the generation of ROS [33]. As shown in Fig. 3A, MCF-7 and MDA-MB-231 cells showed weak fluorescence intensity incubated with Fe $_3\mathrm{O}_4$ @PCBMA due to the insufficient concentration of $\mathrm{H}_2\mathrm{O}_2$ to produce a small amount of ROS. The fluorescence slightly increased after incubation with free SIM, on account of the production of SIM through the MVA pathway. After treatment with



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 ${\rm Fe_3O_4}{\text{-SIM}}$ and ${\rm Fe_3O_4}{\text{@PCBMA-SIM}}$, the intensity of green fluorescence of cells was significantly enhanced. In addition, ${\rm Fe_3O_4}{\text{-SIM}}$ group and ${\rm Fe_3O_4}{\text{@PCBMA-SIM}}$ group exhibit stronger fluorescence intensity than free SIM group from the analysis of flow cytometry (Fig. 3B, C) both in MCF-7 cells and MDA-MB-231 cells, which indicated the enhanced production of ROS in cancer cells induced by ${\rm Fe_3O_4}{\text{-SIM}}$ or ${\rm Fe_3O_4}{\text{@PCBMA-SIM}}$.

In-vitro mechanism of ferroptosis

Statins, a small molecule potent inhibitors of the 3-hydroxy-3-methylglutaryl-coenzyme A reductase (HMGCR), could affect the MVA pathway to control the biosynthesis of cholesterol [34]. This process in suppression of some metabolites and inactivation of GPX4

[12, 35]. For further study the mechanism of ferroptosis induced by Fe₃O₄@PCBMA-SIM, western blot was used to study the MVA pathway. As shown in Fig. 4B, C, the expression of GPX4 and HMGCR protein in MCF-7 cells and MDA-MB-231 cells were influenced by the addition of SIM and Fe₃O₄. For HMGCR protein, both Fe₃O₄@PCBMA and SIM groups all decreased its expression compared with the control group and this lowering effect was obviously in MCF-7 cells (Fig. 4E, G). However, the expression of GPX4 in MCF-7 cells after treated with Fe₃O₄@PCBMA, SIM and Fe₃O₄-SIM were almost no difference and the expression of GPX4 protein decreased obviously after treated with Fe₃O₄@PCBMA and SIM, and further decreased after the addition of Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM in MDA-MB-231 cells, which

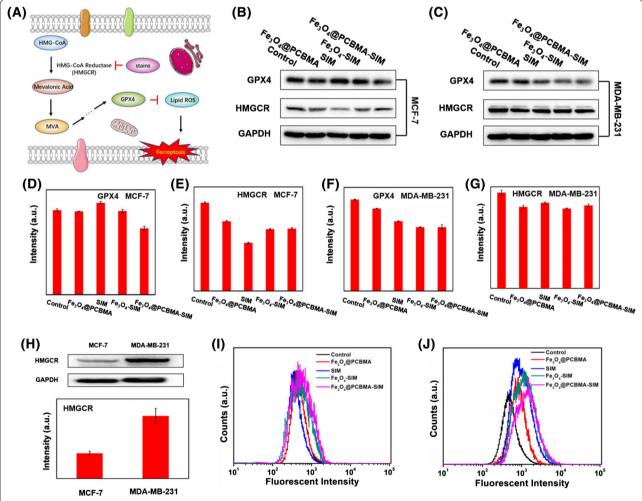


Fig. 4 Schematic illustration of ferroptosis mechanism (**A**); The GPX4 and HMGCR protein expression of MCF-7 cells after treated with Fe₃O₄@ PCBMA, SIM, Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM (**B**, **D**, **E**); The GPX4 and HMGCR protein expression of MDA-MB-231 cells after treatment with Fe₃O₄@PCBMA, SIM, Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM (**C**, **F**, **G**); The HMGCR protein expression of MCF-7 cells and MDA-MB-231 cells and the corresponding intensity (**H**); Flow cytometry analysis of lipid hydroperoxides (LPO) generation in MCF-7 (**I**) and MDA-MB-231 cells (**J**)

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was due to the synergy effect of SIM and $\rm Fe_3O_4$ (Fig. 4D, F). In addition, we have also measurement the amount of HMGCR protein in two cancer cells to study the reason of this phenomenon. As shown in Fig. 4H, the HMGCR protein expression in MDA-MB-231 cells was much higher than in MCF-7 cells, which accounted for the unconspicuous inhibition effect of HMGCR protein. Therefore, the above results show that ferroptosis could occur through MVA pathway to inactivation of GPX4 in TNBC.

Moreover, the inactivation of GPX4 would inhibit the conversion of lipid peroxides into lipid alcohols and the lipid hydroperoxides (LPO) can be used as an important indicator of the ferroptosis [36]. Therefore, the inactivation of GPX4 could promote the accumulation of lipid peroxide level. As shown in Fig. 4I, J, the fluorescence intensity of cells in Fe₃O₄@PCBMA, SIM and Fe₃O₄@ PCBMA-SIM groups showed stronger than that of the control group, which indicated more LPO production after nanodrugs effect. Unlike ROS, the basic LPO in MDA-MB-231 cells was higher than in MCF-7 cells and MDA-MB-231 cells produce more LPO than MCF-7 cells after incubation, which could account for the more cytotoxic of Fe₃O₄@PCBMA-SIM to MDA-MB-231 cells than to MCF-7 cells. Therefore, we could conclude that this nanoplatform could inhibit the expression of HMGCR to downregulate the mevalonate (MVA) pathway and glutathione peroxidase 4 (GPX4), thereby producing more LPO to induce cancer cell ferroptosis, as schematically illustrated in Fig. 4A.

Pharmacokinetics and biodistribution

Nanoparticles could achieve better therapeutic efficacy compared with free drugs due to the high permeability and retention effect of solid tumors [37]. Nevertheless, the unsatisfactory tumor accumulation of nanoparticles is due to the undesirable blood circulation [31]. Therefore, design nanomedicine with prolonged blood circulation property is very important. In this study, we fabricated zwitterionic polymer coating of magnetic nanoparticles for enhancing their blood retention and effectively improved the therapeutic effect of TNBC. MDA-MB-231 tumor-bearing mice were used to study the accumulation property of Fe₃O₄@PCBMA in vivo. After 12, 24 and 48 h injection, visceral organs and tumors were taken out. Then, Fe content were measured using ICP-AES. The accumulation of Fe₃O₄@PCBMA and Fe₃O₄ in tumors at 12 h post injection were $12.6 \pm 2.1\%$ ID/g and $5.2 \pm 1.2\%$ ID/g respectively (Additional file 1: Fig. S7). As shown in Fig. 5A, the accumulation of Fe₃O₄@PCBMA just decreased to $11.1 \pm 4.8\%$ ID/g at the tumor site after 24 h injection and the accumulation of Fe₃O₄ reduced to $3.1 \pm 3.9\%$ ID/g, which indicated that there are more Fe₃O₄@PCBMA intratumor accumulation than Fe₃O₄. Interestingly, the accumulation of Fe₃O₄@PCBMA was still $8.1\pm0.7\%$ ID/g 4 at 48 h post injection (Additional file 1: Fig. S8). Moreover, in addition to the decreased nanoparticles at the tumor site, the residual amounts of nanoparticles in liver and lung also significantly reduced, while the uptake increased in kidney within 24 h, indicating the better metabolic performance of Fe₃O₄@PCBMA.

For further study the nanoparticle retention in blood, pharmacokinetics studies were performed for Fe content measurement using ICP-AES. As shown in Fig. 5B, Fe₃O₄ in blood was decreased to $3.9\pm2.6\%$ ID/mL after 24 h intravenous injection. In contrast, about $15.1\pm4.1\%$ ID/mL of Fe₃O₄@PCBMA were still in blood circulation. Moreover, there was $9.9\pm7.1\%$ ID/mL of Fe₃O₄@PCBMA nanoparticles stayed in blood after 48 h injection. A significant increase indicated that zwitterionic polymer could prolong blood circulation time of nanoparticles in vivo and it also proved the successfully synthesis of Fe₃O₄@PCBMA.

In vivo antitumor combined therapy efficacy

Encouraged by the excellent tumor accumulation of Fe₃O₄@PCBMA, the in vivo antitumor efficacy was then performed on MDA-MB-231 tumor models. All mice were divided into five groups (n=5). As shown in Fig. 5C, there is no obvious change in tumor size between Fe₃O₄@PCBMA group and control group, while tumors grew slower in SIM and Fe₃O₄-SIM than control group. When encapsulated with PCBMA, the growth of tumors suppressed in 22 days and the antitumor rate was 76.1%, which indicated that TNBC could be suppressed under ferroptosis. The most impressive thing is that no significant weight loss was seen in all groups of mice (Fig. 5D), indicating that the safety of Fe₃O₄@PCBMA-SIM. The tumor photos in all groups were consistent with the tumor growth curves. To further study the systemic toxicity of the nanoparticles, all mice were sacrificed at the 22th day treatment and the main organs (liver, spleen, heart, lung, and kidney) were removed and co-stained by hematoxylin and eosin (H&E). As shown in Fig. 6D, there were no obvious tissue damages and noticeable pathological changes in the five groups, which indicated that nanoparticles were possess biosafety and could be an effective nanoplatform for cancer treatment. In addition, TUNEL assay results showed that Fe₃O₄@PCBMA-SIM group had the highest mortality in vivo tumor treatment and it is consistent with its good antitumor inhibition effect (Additional file 1: Fig. S9). In addition, blood biochemistry and routine blood tests were used to tasty the of Fe₃O₄@PCBMA. As shown in Fig. 6A-C, there was no obvious difference of blood indices compared Fe₃O₄

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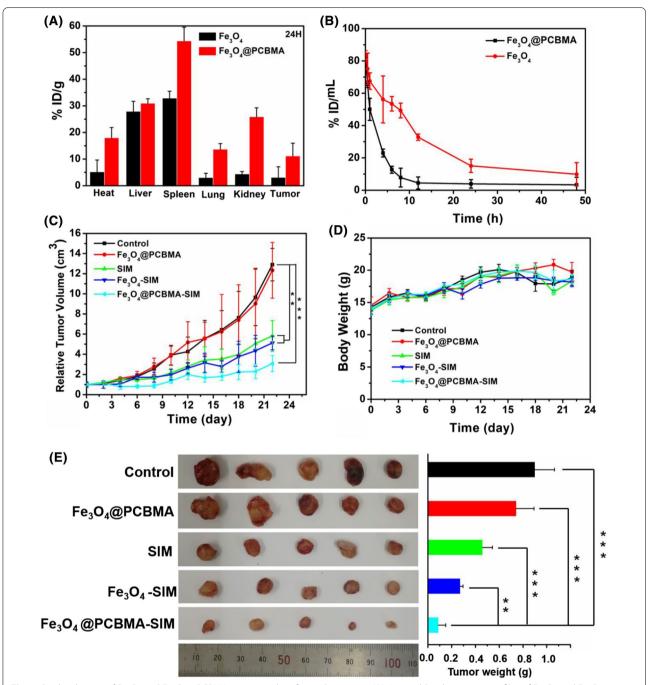


Fig. 5 Biodistribution of Fe_3O_4 and Fe_3O_4 @PCBMA nanoparticles after 24 h injection (**A**); In vivo blood retention profiles of Fe_3O_4 and Fe_3O_4 @ PCBMA nanoparticles (1 mg/mL, 100 μ L) (**B**); Tumor growth curves and body weight changes of nanomedicine. Treatments were performed on day 0, 3, 6 and 9. Error bars indicate the SD (n = 5) (**C**, **D**); Photographs (left) and weights (right) of tumors after therapy (**E**) *P < 0.05 **P < 0.01 ***P < 0.001

and Fe₃O₄@PCBMA group with the control group. Moreover, there is no difference between nanoparticles treatment group and control group of the whole blood panel analysis result (Additional file 1: Fig. S10).

Therefore, $Fe_3O_4@PCBMA$ had good biocompatibility without liver and kidney toxicity.

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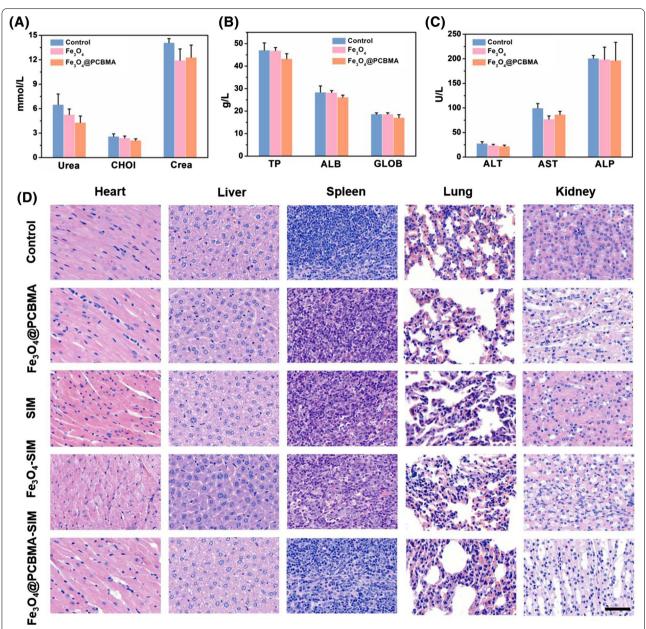


Fig. 6 Blood biochemistry indices of renal and hepatic function including alkaline phosphatase (ALP), alanine aminotransferase (ALT), aspartate aminotransferase (AST) (**A**); globulin (GLOB), total protein (TP), albumin (ALB) (**B**) and creatinine (CREA), cholesterol (CHOL), UREA (**C**); H&E-stained slices of liver spleen, heart, lung, and kidney from each group (**D**), Scale bars were 50 μm

Conclusion

Ferroptosis is a key tumor suppression mechanism. Herein, we presented the ferroptosis nanomedicine by loading simvastatin (SIM), a ferroptosis drugs, into zwitterionic polymer coated of magnetic nanoparticles (Fe $_3$ O $_4$ @PCBMA), thereby improving the therapeutic effect of triple negative breast cancer. This drug

delivery platform was demonstrated to have higher toxicity against MDA-MB-231 than MCF-7, which demonstrated that statins could effectively kill triple negative breast cancer. Furthermore, the western blot result illustrated SIM could through inhibit HMGCR to inhibit the mevalonate pathway and deactivate GPX4. With the inherited blood circulation property and ferroptosis effect, Fe $_3$ O $_4$ @PCBMA-SIM was intravenously injected into MDA-MB-231 tumor-bearing mice and

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their treatment efficiency in vivo was evaluated. Our finding highlights that ${\rm Fe_3O_4@PCBMA\textsc{-}SIM}$ exhibit an excellent tumor suppression, which will open an avenue of triple negative breast cancer therapy.

Materials and methods

Synthesis of magnetic nanoparticles and modification

 ${\rm Fe_3O_4}$ nanoparticles were prepared by the solvothermal reaction method [38]. Typically, ${\rm FeCl_3\cdot 6H_2O}$ (0.54 g, 2 mM), sodium acetate (1.2 g, 15 mM), sodium citrate dihydrate (0.24 g, 1 mM) and 20 mL ethylene glycol mixed and stirred for 30 min. Then transferring the above solution into a three-necked flask (100 mL) under 200 °C for 10 h. Finally, the resulting products were dispersed in 50 mL of deionized water for further use.

Before encapsulating the zwitterionic polymer, 200 mg of ${\rm Fe_3O_4}$, 5 mL of deionized water, 1.5 mL of aqueous ammonia solution (28–30%) was mixed with 35 mL of ethanol and stirred at 60 °C for 30 min. Then, 300 mg of MPS was slowly added for another 12 h. Finally, the obtained products were washed three times with ethanol and were named as ${\rm Fe_3O_4\text{-}MPS}$.

Synthesis of carboxybetaine methacrylate (CBMA) monomer

In a typical recipe [31], 3.14 g of 2-(dimethylamino) ethyl methacrylate was dissolved in 25 mL of anhydrous acetone, to which 1.5 mL of β -propiolactone was added and stirred at 4 °C for 6 h. The CBMA monomer was obtained by extraction filtration and washed with diethyl ether.

Synthesis of Fe₃O₄@PCBMA

 ${\rm Fe_3O_4@PCBMA}$ were prepared by modified reflux-precipitation polymerization method [39]. Using a typical method, 5 mL ethanol, 4 mg of AIBN, 20 mg of ${\rm Fe_3O_4}\text{-MPS}$, 5 mg of BAC, 95 mg of CBMA monomer and 35 mL of acetonitrile were added into a three-necked flask and stirred at 90 °C for 1 h under ${\rm N_2}$. Then the resulting product washed with ethanol and deionized water and obtained the composite nanoparticles, denoted as ${\rm Fe_3O_4@PCBMA}$.

SIM loading of Fe₃O₄@PCBMA (Fe₃O₄@PCBMA-SIM)

Firstly, 3 mg of SIM, 1 mL of deionized water, 2 mL ethanol and 10 mg of ${\rm Fe_3O_4@PCBMA}$ were mixed. The above solution was stirred in the dark for 24 h at room temperature and then vacuum-rotary evaporation to remove solvent and washed with water. Finally, UV–vis spectrophotometer was used to calculate the loading rate of SIM at 238 nm.

SIM releasing from Fe₃O₄@PCBMA-SIM

10 mg of $Fe_3O_4@PCBMA$ -SIM were dispersed into 5 mL of GSH with different concentrations (0, 5, 10 mM). Afterwards, the mixed solution was stirred at 37 °C under dark conditions. After a specific time interval, centrifuge for collecting the supernatant and added another 5 mL of GSH solution. The SIM releasing amount was calculated according to the UV–vis absorbance value of SIM in the supernatant.

The degradation of Fe₃O₄ and Fe₃O₄@PCBMA

To measure the degradation of nanoparticles, Fe_3O_4 and Fe_3O_4 @PCBMA (100 µg/mL) were placed in a 1.4×10^4 Dalton dialysis bag. Afterwards, the dialysis bag was dip in 200 mL of PBS (pH 5.0, 6.5, 7.4) with different concentrations of GSH (0 mM, 10 mM). And then, 2 mL of solution was removed and added 2 mL of PBS with GSH at a certain point in time. ICP-AES was used to measure the amount of Fe.

In vitro cytotoxicity assay

The cytotoxicity of Fe_3O_4 , Fe_3O_4 @PCBMA, free SIM, Fe_3O_4 -SIM and Fe_3O_4 @PCBMA-SIM were measured according to previous report [40]. Firstly, MCF-7 and MDA-MB-231 cells were incubated into 96-well plate. After 12 h incubation, different concentrations of Fe_3O_4 . Fe $_3O_4$ @PCBMA, free SIM, Fe_3O_4 -SIM and Fe_3O_4 @PCBMA-SIM were added respectively. Afterwards, CCK-8 was added after 24 h of culture to measuring the cell viability.

Cell uptake assay

Before cell uptake assay, MCF-7 cells and MDA-MB-231 cells were firstly cultured in a 35-mm glass-bottomed dish. Then SIM, Fe $_3O_4$ @PCBMA, Fe $_3O_4$ -SIM, Fe $_3O_4$ @PCBMA-SIM were added to MCF-7 cells and MDA-MB-231 cells (500 μL , 50 $\mu g/mL$). Nanoparticle untreated cells were used as control group. After culturing 4 h at 37 °C, FerroOrange probe (1 $\mu mol/L$) was added to each dish for 30 min at 37 °C and washed with PBS. Finally, CLSM was used to capture the fluorescence images. (CLSM, Ex: 561 nm, Em: 570–620 nm).

Measuring the reactive oxygen species (ROS) and lipid hydroperoxides (LPO) generation

To investigate the ROS generation, MCF-7 cells and MDA-MB-231 cells were cultured in a 35-mm glass-bottomed dishes. After 24 h incubation, SIM, Fe $_3O_4$ @ PCBMA, Fe $_3O_4$ -SIM, Fe $_3O_4$ @PCBMA-SIM suspension (500 μ L, 50 μ g/mL) were added to each dish and continue cultivation. Then, DCFH-DA (an ROS probe, Ex: 488 nm, Em: 537 nm) was added to each dish after 6 h cultivation for measuring reactive oxygen species (ROS) produced

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by cells. CLSM was used to capture the fluorescence images of cells.

In addition, MCF-7 cells and MDA-MB-231 cells were also seeded in a 6-well plate for the flow cytometry measurement. Similarly, nanoparticles (500 μL , 50 $\mu g/mL)$ were added to 6-well plate. DCFH-DA (ROS probe) and C11-BODIPY (LPO probe) were used to stain above cells. Finally, all cells were digested for flow cytometry measurement.

In vivo biodistribution and pharmacokinetics of $Fe_3O_4@$ PCRMA

Before biodistribution and pharmacokinetics of Fe $_3O_4$ @ PCBMA nanoparticles, a subcutaneous tumor model of MDA-MB-231 cells was firstly constructed. After the tumor volume reached 60 mm 3 , the tumor-bearing nude mice were treated with Fe $_3O_4$ and Fe $_3O_4$ @PCBMA (2 mg/mL) through injected intravenously. After 12 h, 24 h and 48 h, spleen, Heart, kidney, liver, lung and tumor were taken out and dissolved by acid mixture (Vperchloric acid:Vhydrochloric acid=1:4). After diluted with deionized water and filtered through a 0.22 mm membrane, iron distribution in tissues were measured by ICP-AES.

Similarly, eight Balb/c nude mice (male, 20–22 g) were used to measure the pharmacokinetics of Fe $_3$ O $_4$ @ PCBMA. Firstly, Fe $_3$ O $_4$ and Fe $_3$ O $_4$ @PCBMA (2 mg/mL) were intravenously injected into female mice (n=3). Then, the whole blood (30 μ L) was extracted through orbital sinus at different time points (0, 15 min, and 1, 2, 4, 8, 12, 24, 48 h) and dissolved by acid mixture as above for detected iron distribution in blood.

In vivo therapy of ferroptosis

To test the therapy of ferroptosis in vivo, MDA-MB-231 cells were injected to subcutaneous of mice to build a tumor model. When the tumor volume reaches about 60 mm³, all mice were randomly divided into five groups (n=5) for various treatments. Then, mice were treated with PBS, Fe₃O₄@PCBMA, SIM, Fe₃O₄-SIM and Fe₃O₄@PCBMA-SIM through injected intravenously. The injected doses of SIM were 4 mg/kg body weight in each mouse on days 0, 3, 6, and 9. Date was collected every other day for recording the change of body weight and tumor volume after drugs treated. After 22 days of therapy, the organs (liver, kidneys, spleen, lung, heart and tumor) of each group mice were taken out and dispersed in 4% paraformaldehyde for H&E stain. Finally, TUNEL assay was used to measurement the apoptotic cells in the tumor slices.

Blood biochemistry and routine blood testing

To measure the safety of nanomedicine, 12 female ICR mice (25–28 g) were injected intravenously with 100 μL

of PBS, Fe₃O₄ and Fe₃O₄@PCBMA. After 24 h, blood sample from each mouse were taken out and measured the biochemical and routine blood testing indexes to test the safety of nanoparticles.

Statistical analysis

The data were showed as mean \pm standard deviation. The differences between groups were performed by one-way ANOVA with Dunnett's multiple comparisons test. P < 0.05 was considered statistically significant.

Supplementary Information

The online version contains supplementary material available at https://doi.org/10.1186/s12951-021-01058-1.

Additional file 1: Fig. S1. FT-IR spectra of Fe $_3$ O $_4$ and Fe $_3$ O $_4$ -MPS nanoparticles. **Fig. S2.** TEM images of Fe $_3$ O $_4$ @PCBMA-SIM nanoparticles (A); SEM images of Fe $_3$ O $_4$ @PCBMA-SIM nanoparticles (B). **Fig. S3.** DLS diameters of Fe $_3$ O $_4$ @PCBMA-SIM nanoparticles dispersed in water, PBS, bovine serum albumin (BSA) and Dulbecco's Modified Eagle Medium (DMEM). **Fig. S4.** The degradation property of Fe $_3$ O $_4$ nanoparticles dispersed in different concentrations of GSH (0 mM and 10 mM) and pH values (5.0, 6.5 and 7.4). **Fig. S5.** Cell viability of ferrous sulfate to MDA-MB-231 cells. **Fig. S6.** Cell viability of ferrous sulfate to MCF-7 cells. **Fig. S7.** Biodistribution of Fe $_3$ O $_4$ and Fe $_3$ O $_4$ and Fe $_3$ O $_4$ @PCBMA nanoparticles after 12 h injection. **Fig. S8.** Biodistribution of Fe $_3$ O $_4$ and Fe $_3$ O $_4$ and Fe $_3$ O $_4$ @PCBMA nanoparticles after 4b h injection. **Fig. S9.** TUNEL staining of tumor tissues. Scale bars was 20 μ m. **Fig. S10.** Whole blood panel analysis of nanoparticle-treated mice at 24 h post-injection.

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Authors' contributions

XXY, HBP and WLY designed the research. XXY conducted the experiments. XXY, HBP and WLY wrote the manuscript. JT, HBP and WLY analyzed the data and revised the manuscript. XXY, RHX and YBC participated in the animal experiments. All authors read and approved the final manuscript.

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Availability of data and materials

All data generated or analyzed during this study are included in this published article.

Declarations

Ethics approval and consent to participate

All animal experiments were carried out with full authorization approved by the ethical committee for animal care of School of Pharmacy, Fudan University (Approval No. 2014-03-YJ-PZQ-01).

Consent for publication

All authors agree to be published.

Competing interests

The authors declare that they have no competing interests.

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